

GRAPHING TRENDS OF THE PERIODIC TABLE

CHEMISTRY

NAME: _____

DATE: _____

PERIOD: _____

Graphing Periodic Table Trends

Directions: Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Atomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends, (i.e. after atomic number 10). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

Atomic Radius (pm)							
Lithium - 3	152	Sodium - 11	190	Potassium - 19	243	Rubidium - 27	265
Beryllium - 4	112	Magnesium - 12	145	Calcium - 20	194	Strontium - 28	219
Boron - 5	87	Aluminum - 13	118	Gallium - 31	136	Indium - 49	156
Carbon - 6	67	Silicon - 14	111	Germanium - 52	125	Tin - 50	145
Nitrogen - 7	55	Phosphorus - 15	98	Arsenic - 33	114	Antimony - 51	133
Oxygen - 8	48	Sulfur - 16	88	Selenium - 34	103	Tellurium - 52	123
Fluorine - 9	42	Chlorine - 17	79	Bromine - 35	94	Iodine - 53	115

Ionization Energy (kJ/mol)							
Lithium - 3	520	Sodium - 11	496	Potassium - 19	419	Rubidium - 27	403
Beryllium - 4	900	Magnesium - 12	738	Calcium - 20	590	Strontium - 28	550

Atomic Radius



Trends In The Periodic Table Graphing

Michael Buckley



Trends In The Periodic Table Graphing:

Chemistry in the Community American Chemical Society, 2006-01-31 This laboratory based text centres itself around decision making activities where students apply their chemistry knowledge to realistic situations This fifth edition includes more photographs new drawings and new design

Computer Based Projects for a Chemistry Curriculum Thomas J. Manning, Aurora P. Gramatges, 2013-04-04 This e book is a collection of exercises designed for students studying chemistry courses at a high school or undergraduate level The e book contains 24 chapters each containing various activities employing applications such as MS excel spreadsheets and Spartan computational modeling Each project is explained in a simple easy to understand manner The content within this book is suitable as a guide for both teachers and students and each chapter is supplemented with practice guidelines and exercises Computer Based Projects for a Chemistry Curriculum therefore serves to bring computer based learning a much needed addition in line with modern educational trends to the chemistry classroom

Basic Concepts of Chemistry Leo J. Malone, Theodore O. Dolter, 2011-12-27 The 9th edition of Malone's Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter WileyPLUS sold separately from text

Science Units for Grades 9-12 Randy L. Bell, Joe Garofalo, 2005 Sample topics include cell division virtual dissection earthquake modeling the Doppler Effect and more

Jacaranda Core Science Stage 4 New South Wales Australian Curriculum, 3e learnON and Print Paul Arena, 2025-01-28

The Science Teachers Bulletin, 1989

General Science Richard H. Moyer, 1986

Focus on Physical Science Annette Miele Saturnelli, 1989

Chemistry James E. Brady, John R. Holum, 1993-01-21 Offers accurate lucid and interesting explanations of basic concepts and facts of chemistry while helping students develop skills in analytical thinking and problem solving Students are taught in a variety of ways to think of skills as tools that can be used to solve complex problems Several aids are included to help focus and inspire student interest frequent reference to common chemicals in commercial products numerous photographs of reactions in chapter practice exercises following worked examples

Prentice Hall Science Explorer Michael J. Padilla, 2002

The Science Teacher, 1997 SCC Library has 1964 cur

Catalog of Instructional Films (16mm, Super 8, and 8mm) for College Chemistry Advisory Council on College Chemistry, 1969

Physical Science Michael Buckley, 2006-09-01 There's no such thing as too much practice This reproducible program builds skills incrementally By inviting students to show what they

know in a variety of new formats these stimulating lessons will enable struggling students to actually enjoy the learning process As in all of the binder programs the dual emphasis is on 1 mastery of the basics and 2 improved critical thinking

Prentice Hall Chemistry, 2000 ChemCom American Chemical Society, 1993 *Chemistry* Martin Stuart Silberberg, 1996 This general chemistry text centres on the theme that observable change in chemical systems is the result of molecular change The aims of this edition are to enable students to perceive matter and change at the molecular level and to help build student confidence in their ability to solve chemical problems as they discover the relevance of chemistry to their lives Instructional Media and the New Technologies of Instruction Robert Heinich, Michael Molenda, James D. Russell, 1993 Grade level 1 2 3 4 5 6 7 8 9 10 11 12 e i s t **SourceBook Version 2.1**, 1998 Holt Chemistry Ralph Thomas Myers, 2004 **Chemistry; Structure and Reactions** Milton K. Snyder, 1966

Unveiling the Power of Verbal Art: An Emotional Sojourn through **Trends In The Periodic Table Graphing**

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