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1452

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Paper 2

Oct./Nov. 2022

2 hours



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INSTRUCTIONS TO CANDIDATES:

Section A consists of 10 structured questions. Answer all the questions in this section.

Answers to these questions must be written in the spaces provided.

Section B consists of 4 semi-structured questions. Answer any two questions from this section. Answers to the questions must be written in the answer booklet(s) provided.

In both sections all working must be clearly shown and must be in blue or black ink.

Any work done in pencil will not be marked except drawings.

Mathematical tables and silent non-programmable calculators may be used.

Where necessary use:

[$H=1$, $C=12$, $O=16$, $Mg=24$]

1 mole of gas occupies 24 l at room temperature.

1 mole of gas occupies 22.4 l at s.t.p.

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1	2	3	4	5	6	7	8	9	10	11	12	13	14	Total

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Question 8

Uce Questions Chemistry 2013

Francis Jones



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