

Solubility Rules

Soluble compounds include ionic compounds containing ...

1. ... alkali metal ions (such as Na^+ , K^+ , etc.).
2. ... the ammonium ion (NH_4^+).
3. ... chloride, bromide, or iodide ions **EXCEPT** when paired with Ag^+ , Pb^{2+} , or Hg_2^{2+} ions.
4. ... nitrate, acetate, or chlorate ions **EXCEPT** $\text{AgC}_2\text{H}_3\text{O}_2$.
5. ... the sulfate ion **EXCEPT** when paired with Pb^{2+} , Ca^{2+} , Sr^{2+} , Ba^{2+} , or Ra^{2+} ions.

Insoluble compounds include ionic compounds containing ...

6. ... the sulfide ion **EXCEPT** when paired with alkali metal, alkaline earth metal, or ammonium ions.
7. ... the hydroxide ion **EXCEPT** when paired with alkali metal, ammonium, or Sr^{2+} , or Ba^{2+} ions.
8. ... phosphate, carbonate, or sulfite ions **EXCEPT** when paired with alkali metal or ammonium ions.

Ws15 1 Solubility Rules

Annelies Wilder-Smith

A red circular graphic with a gradient, appearing as a semi-circle or a partial circle, located to the right of the author's name.

Ws15 1 Solubility Rules:

Dictionary Catalog of the Department Library United States. Department of the Interior. Library, 1967 Science
Citation Index , 1975 Vols for 1964 have guides and journal lists

Ws15 1 Solubility Rules Book Review: Unveiling the Magic of Language

In a digital era where connections and knowledge reign supreme, the enchanting power of language has become more apparent than ever. Its ability to stir emotions, provoke thought, and instigate transformation is really remarkable. This extraordinary book, aptly titled "**Ws15 1 Solubility Rules**," published by a highly acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound affect our existence. Throughout this critique, we will delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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