

Annotated periodic table of the elements

Group names*

- 1 Hydrogen & the alkali metals
- 2 Alkaline earth metals
- 3 Coinage metals
- 12 Volatile metals
- 13 Borogens
- 14 Chalcogens
- 15 Pnictogens
- 16 Chalcogens
- 17 Halogens
- 18 Helium group

*Groups 3 to 10 are named after their first members i.e. Group 3 is the Scandium group

1	2	3	4	5	6	7	8	9	10	11	12	13	14	15	16	17	18
1 H 1.01 Hydrogen												B 10.81 Boron	C 12.01 Carbon	N 14.01 Nitrogen	O 16.00 Oxygen	F 18.99 Fluorine	Ne 20.18 Neon
3 Li 6.94 Lithium	4 Be 9.012 Beryllium											Al 26.98 Aluminum	Si 28.09 Silicon	P 30.97 Phosphorus	S 32.06 Sulfur	Cl 35.45 Chlorine	Ar 39.95 Argon
5 Na 22.99 Sodium	6 Mg 24.31 Magnesium											K 39.10 Potassium	Ca 40.08 Calcium				
7 K 39.10 Potassium	8 Ca 40.08 Calcium											Rb 85.47 Rubidium	Sr 87.62 Strontium				
9 Rb 85.47 Rubidium	10 Sr 87.62 Strontium											Cs 132.91 Cesium	Ba 137.33 Barium				
11 Cs 132.91 Cesium	12 Ba 137.33 Barium											Fr 223.02 Francium	Ra 226.03 Radium				
13 Fr 223.02 Francium	14 Ra 226.03 Radium																

Most common oxidation states, electronegativity (Allen-Pauling)

Atomic number and symbol, atomic weight and name

57 to 70

89 to 102

19 Sc 44.96 Scandium	20 Ti 47.87 Titanium	21 V 50.94 Vanadium	22 Cr 52.00 Chromium	23 Mn 54.94 Manganese	24 Fe 55.85 Iron	25 Co 58.93 Cobalt	26 Ni 58.69 Nickel	27 Cu 63.55 Copper	28 Zn 65.38 Zinc	29 Ga 69.72 Gallium	30 Ge 72.64 Germanium	31 As 74.92 Arsenic	32 Se 78.96 Selenium	33 Br 79.90 Bromine	34 Kr 83.80 Krypton
35 Y 88.91 Yttrium	36 Zr 91.22 Zirconium	37 Nb 92.91 Niobium	38 Mo 95.94 Molybdenum	39 Tc 98.91 Technetium	40 Ru 101.07 Ruthenium	41 Rh 101.07 Rhodium	42 Pd 106.32 Palladium	43 Ag 107.87 Silver	44 Cd 112.41 Cadmium	45 In 114.82 Indium	46 Sn 118.71 Tin	47 Sb 121.76 Antimony	48 Te 127.60 Tellurium	49 I 126.91 Iodine	50 Xe 131.29 Xenon
51 La 138.91 Lanthanum	52 Ce 140.12 Cerium	53 Pr 140.91 Praseodymium	54 Nd 144.24 Neodymium	55 Pm 144.91 Promethium	56 Sm 150.36 Samarium	57 Eu 151.96 Europium	58 Gd 157.25 Gadolinium	59 Tb 158.93 Terbium	60 Dy 162.50 Dysprosium	61 Ho 164.93 Holmium	62 Er 167.26 Erbium	63 Tm 168.93 Thulium	64 Yb 173.05 Ytterbium	65 Lu 174.97 Lutetium	66 Hf 178.49 Hafnium
67 Ta 180.95 Tantalum	68 W 183.84 Tungsten	69 Re 186.21 Rhenium	70 Os 190.23 Osmium	71 Ir 192.22 Iridium	72 Pt 195.08 Platinum	73 Au 196.97 Gold	74 Hg 200.59 Mercury	75 Tl 204.38 Thallium	76 Pb 207.2 Lead	77 Bi 208.98 Bismuth	78 Po 209 Polonium	79 At 210 Astatine	80 Rn 222 Radon	81 Fr 223 Francium	82 Ra 226 Radium
83 Ac 227.03 Actinium	84 Th 232.04 Thorium	85 Pa 231.04 Protactinium	86 U 238.03 Uranium	87 Np 237.05 Neptunium	88 Pu 244.06 Plutonium	89 Am 243.06 Americium	90 Cm 247.07 Curium	91 Bk 247.07 Berkelium	92 Cf 251.08 Californium	93 Es 252.08 Einsteinium	94 Fm 257.10 Fermium	95 Md 258.10 Mendelevium	96 No 259 Nobelium	97 Lr 260.10 Lawrencium	98 Og 284.10 Oganesson

Radioactive

Approximate melting point in °C

CLASSIFICATION	Metals	Nonmetals
Active	Pre-transition Alkali earth Alkaline	Corrosive
Transition	Intermediate	
Frontier	Metalloid	
Node (Ru-Pt, Os-Au)	Node gas	

67 La 138.91 Lanthanum	68 Ce 140.12 Cerium	69 Pr 140.91 Praseodymium	70 Nd 144.24 Neodymium	71 Pm 144.91 Promethium	72 Sm 150.36 Samarium	73 Eu 151.96 Europium	74 Gd 157.25 Gadolinium	75 Tb 158.93 Terbium	76 Dy 162.50 Dysprosium	77 Ho 164.93 Holmium	78 Er 167.26 Erbium	79 Tm 168.93 Thulium	80 Yb 173.05 Ytterbium	81 Lu 174.97 Lutetium
83 Ac 227.03 Actinium	84 Th 232.04 Thorium	85 Pa 231.04 Protactinium	86 U 238.03 Uranium	87 Np 237.05 Neptunium	88 Pu 244.06 Plutonium	89 Am 243.06 Americium	90 Cm 247.07 Curium	91 Bk 247.07 Berkelium	92 Cf 251.08 Californium	93 Es 252.08 Einsteinium	94 Fm 257.10 Fermium	95 Md 258.10 Mendelevium	96 No 259 Nobelium	97 Lr 260.10 Lawrencium

Metals Nonmetals

Active metals

Pre-transition
Light, reactive, mostly of low mechanical strength, melting and boiling points. The group 1 members are easily cut with a knife. Only Be, Mg and Al have any structural uses. Most form basic oxides (Be and Al are amphoteric).

Alkali earth
Closely resemble Ca, Sr and Ba, but many are heavier, and they form mostly pale-colored compounds. Similar to one another, esp. Li to Cu (the lanthanides), and hard to separate for that reason. Abundant but widely dispersed; commercially viable concentrations are rare. Oxides are strongly to moderately basic. Sc, Yb and Lu exhibit weak amphotericism.

Actinides
Soft, dense and reactive metals, those from Pa onwards are synthetic. Early actinides (Th-Am) show some similarities to transition metals and have basic, amphoteric, or acidic oxides; late actinides are more like lanthanides. Many actinides form colored compounds.

Transition metals
Moderate to high density metals, with high melting and boiling points, many have high hardness and mechanical strength. Those from Rf onwards are synthetic. Chemically show variable valency and strong tendency to form coordination and colored compounds. Oxides are basic, amphoteric or acidic, depending on oxidation state.

Noble metals
Resistant to corrosion and oxidation, unlike base metals such as Fe, Ni, Cu and Zn. Commonly found together in native form. Good catalyst properties. Very weakly electropositive; inert to oxides.

Frontier metals
Soft (or brittle), low strength, and with melting points lower than transition metals. Crystalline structures tend to show directional bonding, with generally greater complexity or lower nearest neighbours than other metals. All used or tried as soft metal lubricants. Most are chemically weak; a minority are moderately electropositive. Characterized, to varying degrees, by covalent bonding tendencies, amphotericism, and anionic species such as zincates, stannates, and bismuthates.

Metalloids
Metallic looking brittle, fair electrical conductors. All known in semiconducting forms. Moderate ionization energies (750–1,000 kJ/mol), and electronegativity values (1.8–2.3). Mostly behave chemically as non-metals. Can form alloys with metals. Amphoteric or weakly acidic oxides.

Intermediate nonmetals
N and P are colorless gases, C, Si, and Se are metalloids looking solids while S is yellow. The latter four are brittle but all known in malleable, plastic or ductile forms. Moderate to high ionization energies, low to high electron affinities, moderate to high electronegativity values, relatively poor to moderately strong oxidizing agents, tend to form covalent or metallic to near-metallic compounds with metals. Their oxides are acidic, or neutral (P₂O₃, CO, NO, N₂O).

*as black P, the most stable form in ambient conditions

Group 1
This starts as Li, H, and then bifurcates into a La branch i.e. La-Ac, and a Lu branch i.e. Lu-Lr. Trends going down: Li-H, and then La branch, resemble those of groups 1–2.

*Lu branch, more resemble those of groups 1–2

Corrosive nonmetals
O, F, and Cl are, respectively, colorless, pale yellow, and pale yellow-green gases. Br is a red-brown liquid, I is a metallic looking solid. Chemically tend to have to high ionization energies, electron affinities, and electronegativity values, and be relatively strong oxidizing agents. Manifestations include oxygen's major association with the ubiquitous processes of corrosion and combustion, and the intrinsically corrosive nature of the non-metallic halogens. All five tend to form predominantly ionic compounds with metals. Their oxides are acidic.

Noble gases
Colorless, colorless, non-flammable gases with very low chemical reactivity. The first noble gas compound, of approximate composition XePF₆, was not prepared until 1962. More than half-a-thousand noble gas compounds were synthesized in the ensuing half-century. Compounds of He, Ne, and Ar, in ambient conditions, are not known.

Unscrambling The Periodic Table Answers

DP Hallahan



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Microsoft Word Simple Projects Jan Rader, Jan Ray, 2001-04 Projects for language arts social studies science and math Provided templates can be modified to meet specific needs Project samples also provided

Scientifica, 2005 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities

Scientifica Essentials, 2005

Unscramble Your Nest Egg John J. Cunningham, 2013-10-25 Pay off mortgage by age 50 55 deposit former mortgage payments into income earning accounts and result will be a cash nest egg big enough to pay 50 percent of living expenses during 20 years of retirement

Periodic Table Cindy Blobaum, 2005 What do chocolate chip cookies chemistry and logic have in common They are the basis for a unit that lets students become actively engaged in discovering the arrangement of the periodic table This learning activity takes the periodic table out of the static presentation usually associated with textbooks and chemistry courses and interjects an element of discovery The two activities in this unit provide students with information that they have to arrange in organized charts In the process of creating the arrangements students will be involved in problem solving and will gain an appreciation for the scientific process of exploration and verification This dynamic unit meets national science standards in seven teaching and content areas Bring the periodic table to life with this hands on minds on unit Book jacket

Scientifica Pupil Book 8 (Levels 4-7) David Sang, Lawrie Ryan, Jane Taylor, Peter Ellis, 2005 Bring your science lessons to life with Scientifica Providing just the right proportion of reading versus doing these engaging resources are differentiated to support and challenge pupils of varying abilities

New Modular Science for GCSE. Ann Fullick, 1998 This homework book provides graded homework questions to help the teacher set homework for a range of abilities It offers 14 sets of homework questions for each of the 12 modules in the NEAB science syllabus

The Science Teacher, 2008

Cyber Science 6 Tm' 2007 Ed., *Story of the World Activity Book 4 Modern Age* Susan Wise Bauer, 2005-11-29 Presents a history of the ancient world from 6000 B C to 400 A D

Verbal Reactions - Word Scrambles With a Chemical Flavor (Easy) Chris McMullen, Carolyn Kivett, 2011-12-09

VERBAL ReAcTiONS are a new form of word scrambles with a chemical flavor You don't need to know any science to be able to solve VERBAL ReAcTiONS puzzles but the puzzles bear a resemblance to chemical reactions

EXAMPLE Here is a sample

Es 2 S P Si 2 O N _ _ _ _ _ _ _ _ This VERBAL ReAcTiON is a word scramble consisting of one Es two S s one P one Si two O s and one N That is the word scramble contains the elements Es S S P Si O O and N Unscramble these elements to form an 8 symbol word that's why there are 8 blanks in the puzzle

ANSWER For this puzzle the answer is P O S S Es Si O N

possession These VERBAL ReAcTiONS resemble chemical reactions in two ways

First the scrambled elements appear added together on the left of the reaction with coefficients like the number 2 in the puzzle above telling you how many of each element the solution contains and you fill in the result of the VERBAL ReAcTiON by rearranging the elements and writing them on the blanks on the right side of the reaction

Secondly all of the

solutions are chemical words A chemical word is a word that can be made using symbols from the periodic table For example the chemical word POSSESiON is made using the symbols for phosphorus P oxygen O sulfur S Einsteinium Es silicon Si and nitrogen N You don t need to be familiar with the periodic table to solve these problems nor do you need to know any chemistry You just need to be able to count and unscramble elements to make words This Easy volume consists of words with 4 to 5 symbols which involves familiarity with common 4 to 10 letter words Other Medium and Hard volumes consist of longer words A unique feature of this book is that there is a Hints section at the back separate from the Answers section for puzzlers who may be stuck and want to check just the first letter of the solution MORE EXAMPLES 1 S Ni Ge U _ _ _ _ 2 2 C N 2 I P _ _ _ _ _ 3 Ti C Cr P Y _ _ _ _ _ 4 2 C U 2 S Es _ _ _ _ _ You can find the answers below Note that this easy volume consists of chemical words with 4 to 5 symbols We recommend starting with our easy puzzles before tackling the medium or hard puzzles available in separate volumes ANSWERS 1 GeNiUS 2 PICNIC 3 CrYPTiC 4 SUCCEsS

Jumpstarters for Properties of Matter, Grades 4 - 12 Eric T. Olson, 2008-12-19 Engage students in describing and comparing general properties of objects like size shape and temperature Then move on to specific types of matter that have characteristic properties such as density and the melting point And lastly have students focus on the notions of elements and compounds Each reproducible page includes five exercises that can be used as whole worksheets for homework assignments and extra practice or cut apart for daily warm up activities Supports NSE standards **Amazing People: Scientists and Mathematicians** Elise Craver, Malaski, Malay, Hailey Scragg, 2023-01-23 Inspirational 1st 2nd and 3rd grade workbooks for kids ages 7 Motivate and inspire children with the Amazing People Scientists and Mathematicians Activity Book 1st 2nd and 3rd grade books are a great way for children to learn about inspirational people who have made important contributions to science and math through a variety of activities that are both fun AND educational Why You ll Love This 1st Grade Third Grade Workbook Engaging and educational activities and wall d cor The activity book features biographies activities and flash cards covering 53 inspirational mathematicians and scientists The activity book also includes 1 inspirational poster great for both homeschool and classroom curriculum and classroom d cor Tracking progress along the way Test your child s knowledge with the flash cards that cover scientists and mathematicians learned throughout the book Use the stickers to reward students on a job well done after completing activities Practically sized for every activity The 256 page 1st grade workbook is sized at 7 75 x 10 625 giving your child plenty of space to complete each exercise About Carson Dellosa For more than 45 years Carson Dellosa has provided solutions for parents and teachers to help their children get ahead and exceed learning goals Carson Dellosa supports your child s educational journey every step of the way The Grade 1 Grade 3 Workbook Contains Biographies and activities covering 53 people 64 flash cards 53 stickers 1 motivational poster **Verbal Reactions - Word Scrambles With a Chemical Flavor (Hard)** Chris McMullen, Carolyn Kivett, 2011-06-18 VERBAL ReAcTiONS are a new form of word scrambles with a chemical flavor You don t need to know any science to be able to solve

VERBAL ReAcTiONS puzzles but the puzzles bear a resemblance to chemical reactions Here is a sample Es 2 S P Si 2 O N ____
 ____ This VERBAL ReAcTiON is a word scramble consisting of one Es two S s one P one Si two O s and one N
 That is the word scramble contains the elements Es S S P Si O O and N Unscramble these elements to form an 8 symbol word
 that s why there are 8 blanks in the puzzle For this puzzle the answer is P O S S Es Si O N possession These VERBAL
 ReAcTiONS resemble chemical reactions in two ways First the scrambled elements appear added together on the left of the
 reaction with coefficients like the number 2 in the puzzle above telling you how many of each element the solution contains
 and you fill in the result of the VERBAL ReAcTiON by rearranging the elements and writing them on the blanks on the right
 side of the reaction Secondly all of the solutions are chemical words A chemical word is a word that can be made using
 symbols from the periodic table For example the chemical word POSSEsSiON is made using the symbols for phosphorus P
 oxygen O sulfur S Einsteinium Es silicon Si and nitrogen N You don t need to be familiar with the periodic table to solve these
 problems nor do you need to know any chemistry You just need to be able to count and unscramble elements to make words
 This Hard volume consists of words with 7 to 8 symbols which involves familiarity with common 8 to 14 letter words Other
 Medium and Easy volumes consist of shorter words A unique feature of this book is that there is a Hints section at the back
 separate from the Answers section for puzzlers who may be stuck and want to check just the first letter of the solution MORE
 EXAMPLES 1 S Ni Ge U ____ 2 2 C N 2 I P ____ 3 Ti C Cr P Y ____ 4 2 C U 2 S Es ____ You
 can find the answers at the end of this paragraph Note that this hard volume consists of chemical words with 7 to 8 symbols
 which are longer than the examples shown here We recommend starting with our easy or medium puzzles before tackling
 these hard puzzles available in separate volumes ANSWERS 1 GeNiUS 2 PICNIC 3 CrYPTiC 4 SUCCEsS *KIRIBATI*
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 PREPARE EFFECTIVELY **Science Reporter** ,2001 **Science News-letter** ,1962-07 Grand Obsession Rosalyn
 Pflaum,1989 A portrait of the scientist and her family winners of three Nobel prizes in two generations **Standard**

Catalog for High School Libraries H.W. Wilson Company, 1993 Each vol is divided into 2 parts 1st 7th ed Dictionary catalog and Classified catalog 8th 9th ed have 3rd part Directory of publishers Popular Electronics , 1978

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web jul 9 2008 i believe there are only about 6 7 heavy guage wires and 2 3 three lighter guage 2 major wiring groupings in

the column and the smaller of the 2 has the heavy gauge wires for starting and i don t think you even need to mess with the other grouping

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