

Year 7 Chemistry Revision

Topic 1: Matter

1. The table below shows the melting points and boiling points of four substances, P, Q, R and S.

Substance	Melting Point (°C)	Boiling Point (°C)
P	0	100
Q	5	80
R	20	90
S	40	120

At which temperature(s) would all the substances be in the same state?

A: 3°C

B: 45°C

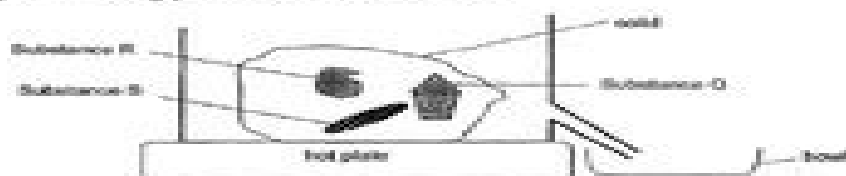
C: 100°C

D: 150°C

- a. B only
- b. A and C only
- c. B and D only
- d. A, C and D only

[1]

2. Jack placed a solid made of substances Q, R and S in the set-up as shown below. Substances Q and R had higher melting points than substance S.



Before switching on the hot plate, he wanted to make sure that the liquid in the bowl contains only substance S. What temperature should he set for the hot plate?

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