Thermochemical Equations

 Thermochemical equations are balanced chemical equations that include the physical states of all reactants and products and the energy change.

•
$$2H_2O(1) \rightarrow 2H_2(g) + O_2(g) \Delta H = 572 \text{ kJ}$$

• $CH_4(g) + 2O_2(g) \rightarrow CO_2(g) + 2H_2O(g)$ $\Delta H = -802 \text{ kJ}$

Thermochemistry Calculations

JE Gale

Thermochemistry Calculations:

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web oct 31 2023 an introduction to ecology ecology is the study of the interactions of living organisms with their environment within the discipline of ecology researchers work at four specific levels sometimes discretely and sometimes with overlap these levels are organism population community and ecosystem