

$$\text{pH} = -\log[\text{H}_3\text{O}^+]$$



Unit 14: Acids & Bases

pH 0 1 2 3 4 5 6 7 8 9 10 11 12 13 14

Increasing acidity pH < 7

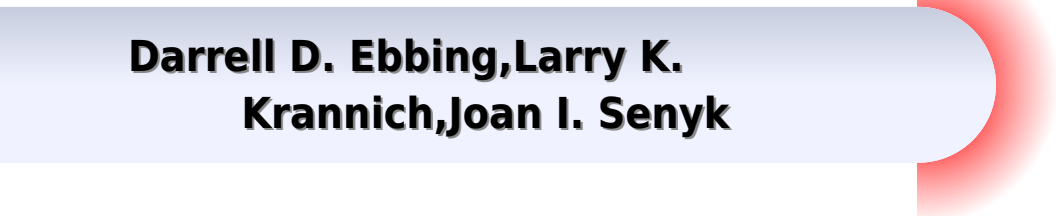
Neutral

Increasing basicity pH > 7



Unit 14 Acids And Bases Answers

**Darrell D. Ebbing, Larry K.
Krannich, Joan I. Senyk**



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