

ACS Physical Chemistry Thermochemistry Study Guide Verified Questions And Answers

What is the third law of thermodynamics? ☒ The entropy of a perfect crystal is zero.

What is the second law of thermodynamics? ☒ As a whole, the universe always tends towards increasing entropy.

What is the relationship between number of microstates and entropy? ☒ More microstates = higher entropy

What is the relationship between temperature and entropy? ☒ Temperature increases, entropy increases (direct)

What is the relationship between atomic weight and entropy? ☒ Heavier = increased entropy (direct)

What happens to entropy when a substance is dissolved in another? ☒ Entropy increases

What is the relationship between moles of gas and entropy? ☒ When moles of gas increase, entropy increases (direct)

What is the relationship between free energy, enthalpy, and entropy?
☒ $\Delta G(\text{system}) = \Delta H(\text{system}) - T\Delta S(\text{system})$

What does change in G measure? ☒ The extent of the spontaneity of a process, and the useful energy available from it

What is the relationship between work and spontaneous processes? ☒ Spontaneous processes produce work

What is the relationship between work and non-spontaneous processes? ☒ Non-spontaneous processes require work

What does -w mean? ☒ Work is being produced - spontaneous

What does a positive sign for w mean? ☒ Work is required - non-spontaneous

Can a change be spontaneous in both directions? ☒ No

What is an extensive property? ☒ Value depends on the amount of substance

Thermochemistry Question Answer Guide

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Patrick E. McMahon, Rosemary McMahon, Bohdan Khomtchouk, 2019-02-13 This work evolved over thirty combined years of teaching general chemistry to a variety of student demographics The focus is not to recap or review the theoretical concepts well described in the available texts Instead the topics and descriptions in this book make available specific detailed step by step methods and procedures for solving the major types of problems in general chemistry Explanations instructional process sequences solved examples and completely solved practice problems are greatly expanded containing significantly more detail than can usually be devoted to in a comprehensive text Many chapters also provide alternative viewpoints as an aid to understanding Key Features The authors have included every major topic in the first semester of general chemistry and most major topics from the second semester Each is written in a specific and detailed step by step process for problem solving whether mathematical or conceptual Each topic has greatly expanded examples and solved practice problems containing significantly more detail than found in comprehensive texts Includes a chapter designed to eliminate confusion concerning

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