

REPORT SHEET

Solubility and Thermodynamics

EXPERIMENT

34

DETERMINING THE K_{sp} OF Ca(OH)_2 AT ROOM TEMPERATURE

Temperature

20.5°C

Concentration of HCl solution

0.0172 M

Volume of HCl used

TRIAL 1

9.30 mL

TRIAL 2

9.40 mL

TRIAL 3

9.32 mL

Moles of OH^-

1.60×10^{-4} mol

1.62×10^{-4}

1.60×10^{-4}

Volume of sample

5.00

5.00

5.00

$[\text{OH}^-]$

$[\text{Ca}^{2+}]$

K_{sp}

Experimentally determined value of K_{sp} for trial 1

1.6×10^{-5}

+

Standard deviation

$\Delta G^\circ =$

Thermodynamics Lab Answers

**Xenia Klinger, Canadian Waterfront
Resource Centre, Royal Commission on
the Future of the Toronto Waterfront
(Canada)**

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