

Unit 4.8 STOICHIOMETRY part 2

(Show all work using dimensional analysis)

1. 2 eggs + 1 egg → 3 soft-boiled

a) If you have 12 eggs of hard and 3 eggs, how many dozens of soft-boiled you make?

Ans: _____

b) What ingredient do you run out of first (limiting reagent)?

Ans: _____

2. 2 Al + 6 HCl → 2 H₂ + 2 AlCl₃

a) If you start with 200 g of aluminum and 750 g of HCl, how many grams of H₂ can be made?

Ans: _____

b) What is the limiting reagent?

Ans: _____

c) Suppose you only make 100 g of H₂. What is your % yield?

Ans: _____

3. 2 Fe₂O₃ + 3 C → 4 Fe + 3 CO₂

a) How many grams of iron can be made from 100 g of Fe₂O₃ and 12.7 g of C?

Ans: _____

b) What is the limiting reagent?

Ans: _____

c) After the above reaction, you produce 25.0 g of Fe. What is your % yield?

Ans: _____

4. 6 CO₂ + 6 H₂O → C₆H₁₂O₆ + 6 O₂

a) How many grams of sugar can be made from 500 g of CO₂ and 500 g of water?

Ans: _____

b) What is the limiting reagent?

Ans: _____

c) What is the excess reagent?

Ans: _____

BONUS: How many grams of three extra reagent are leftover?

How much of the limiting reagent is left?

Ans: _____

How much of the limiting reagent is left? 0.00 0.01 0.02 0.03 0.04 0.05 0.06 0.07 0.08 0.09 0.10

0.11 0.12 0.13 0.14 0.15 0.16 0.17 0.18 0.19 0.20

Ws 44 Stoichiometry Part 2 Answers

IM Harris



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