Thermochemical Equations

 Thermochemical equations are balanced chemical equations that include the physical states of all reactants and products and the energy change.

•
$$2H_2O(1) \rightarrow 2H_2(g) + O_2(g) \Delta H = 572 \text{ kJ}$$

• $CH_4(g) + 2O_2(g) \rightarrow CO_2(g) + 2H_2O(g)$ $\Delta H = -802 \text{ kJ}$

Thermochemistry Calculations

Joseph William Richards

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Metallurgical Calculations: Introduction, Chemical and thermal principles, problems in combustion Joseph William Richards,1906 Metallurgical Calculations Joseph William Richards,1906 Rate Constant Calculation for Thermal Reactions Herbert DaCosta, Maohong Fan, 2011-12-28 Providing an overview of the latest computational approaches to estimate rate constants for thermal reactions this book addresses the theories behind various first principle and approximation methods that have emerged in the last twenty years with validation examples It presents in depth applications of those theories to a wide range of basic and applied research areas When doing modeling and simulation of chemical reactions as in many other cases one often has to compromise between higher accuracy higher precision approaches which are usually time consuming and approximate lower precision approaches which often has the advantage of speed in providing results This book covers both approaches It is augmented by a wide range of applications of the above methods to fuel combustion unimolecular and bimolecular reactions isomerization polymerization and to emission control of nitrogen oxides An excellent resource for academics and industry members in physical chemistry chemical engineering and related fields Technical Report - Jet Propulsion Laboratory, California Institute of Technology, Jet Propulsion Laboratory (U.S.),

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