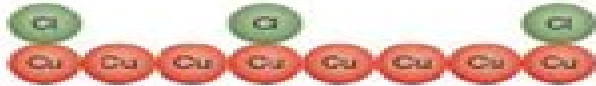


Chemie Overal 3 havo 7^e editie
Uitwerkingen van de RTTI toets A van hoofdstuk 3

Opgave	Antwoord	Score	RTTI																												
1	Harde schijf 1 <table><tr><th>naam</th><th>symbool</th></tr><tr><td>aluminium</td><td>Al</td></tr><tr><td>magnesium</td><td>Mg</td></tr><tr><td>koper</td><td>Cu</td></tr><tr><td>silicium</td><td>Si</td></tr><tr><td>zink</td><td>Zn</td></tr><tr><td>nikkel</td><td>Ni</td></tr><tr><td>fosfor</td><td>P</td></tr><tr><td>kobalt</td><td>Co</td></tr><tr><td>ijzer</td><td>Fe</td></tr><tr><td>chromium</td><td>Cr</td></tr><tr><td>ruthenium</td><td>Ru</td></tr><tr><td>neodymium</td><td>Nd</td></tr><tr><td>boor</td><td>B</td></tr></table> 2 Chloor: niet-metaal 3 Chloor: Cl 4 $-196\text{ }^{\circ}\text{C} = 77\text{ K}$ $77\text{ K} < 172\text{ K}$ dus vast 5 Een laagje koperatomen getekend, eventueel als één geheel of met symbolen. 	naam	symbool	aluminium	Al	magnesium	Mg	koper	Cu	silicium	Si	zink	Zn	nikkel	Ni	fosfor	P	kobalt	Co	ijzer	Fe	chromium	Cr	ruthenium	Ru	neodymium	Nd	boor	B	5 1 1 2 2	T1 T1 T1 T2 I
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kobalt	Co																														
ijzer	Fe																														
chromium	Cr																														
ruthenium	Ru																														
neodymium	Nd																														
boor	B																														
2	Vlamvertragers 6 B 7 Het betekent dat de broomverbindingen bijvoorbeeld door regen niet gemakkelijk vrijkomen uit de behandelde materialen. Daardoor komen er minder broomverbindingen in het milieu. of Bij bluswerkzaamheden komen de broomverbindingen uit niet verbrande materialen via het bluswater niet in het milieu. 8 Het broomatoom bevat 35 protonen en $79 - 35 = 34$ neutronen in de kern en 35 elektronen in de elektronenwolk. 9 De protonen zijn positief geladen en de elektronen zijn negatief geladen. 10 C: de modelvoorstelling geeft elementen en verbindingen weer.. 11 $\text{C}_{12}\text{OBr}_{10}$ 12 C: $12 \times 12,0\text{ u} = 144\text{ u}$ O: $1 \times 16,0\text{ u} = 16,0\text{ u}$ Br: $10 \times 79,9\text{ u} = 799\text{ u}$ Molecuulmassa is $144\text{ u} + 16,0\text{ u} + 799\text{ u} = 959\text{ u}$ $\frac{799}{959} \times 100\% = 83,3\%$ Massapercentage broom is 83,3%	1 2 2 1 1 2 3	T1 I T1 R T2 T1 T1																												

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A Gutmann



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