

AMPLIFY SCIENCE – MIDDLE SCHOOL

THERMAL ENERGY

KEY CONCEPTS NOTES

THERMAL ENERGY KEY CONCEPTS

Unit Question: Why do things change temperature?

Concept 1: Thermal energy is the total kinetic energy of all the particles in an object.

1. What is thermal energy?

2. How does thermal energy relate to temperature?

3. How does thermal energy relate to heat?

4. How does thermal energy relate to work?

5. How does thermal energy relate to the laws of thermodynamics?

Concept 2: Heat is the transfer of thermal energy between two objects.

1. What is heat?

2. How does heat relate to temperature?

3. How does heat relate to work?

4. How does heat relate to the laws of thermodynamics?

5. How does heat relate to the conservation of energy?

Concept 3: Temperature is a measure of the average kinetic energy of the particles in an object.

1. What is temperature?

2. How does temperature relate to thermal energy?

3. How does temperature relate to heat?

4. How does temperature relate to the laws of thermodynamics?

5. How does temperature relate to the conservation of energy?

THERMAL ENERGY GLOSSARY

Adiabatic: A process in which no heat is exchanged between a system and its surroundings.

Boiling: The process of a liquid changing into a gas.

Change: A process in which a system's state or properties change.

Conservation: The principle that the total amount of a quantity remains constant.

Energy: The ability to do work or cause change.

Expansion: The process of a substance increasing in volume.

Heat: The transfer of thermal energy between two objects.

Heat capacity: The amount of heat required to raise the temperature of a substance by one degree.

Heat engine: A device that converts heat into work.

Heat transfer: The process of energy being transferred from one object to another.

Heat work: The process of heat being converted into work.

Internal energy: The total energy of a system, including kinetic and potential energy.

Law of thermodynamics: A principle that governs the behavior of energy and matter.

Pressure: The force exerted by a gas on the walls of its container.

Process: A series of steps or events that lead to a particular outcome.

Temperature: A measure of the average kinetic energy of the particles in a substance.

Thermal energy: The total kinetic energy of all the particles in an object.

Thermal expansion: The process of a substance increasing in volume due to heating.

Thermal energy key concepts: The principles that govern the behavior of thermal energy.

Thermal energy glossary: A collection of terms and definitions related to thermal energy.

Thermal energy notes: A collection of notes and concepts related to thermal energy.

Thermal energy word wall: A collection of words and phrases related to thermal energy.

GREAT ALTERNATIVE TO

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Thermal Energy Notetaking Guide

Rachel S Tattersall



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