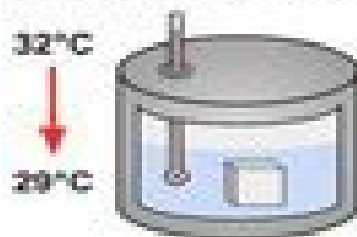


Instructions: Read each scenario and answer the questions below about calorimetry.



A piece of metal is placed in a calorimeter.

The temperature reading on the thermometer decreases from 32°C to 29°C.

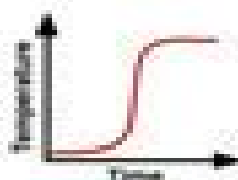
- 19) Which substance underwent an exothermic process? _____
- 20) Which substance underwent an endothermic process? _____
- 21) Heat flowed from the _____ into the _____

Calorimeter Readings	Temperature
Initial Temperature of Steel	150°C
Initial Temperature of Water	95°C
Final Temperature of Water	98°C

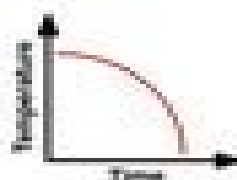
A cube of steel is placed in a calorimeter filled with water.

Initial and final temperatures are shown in the table.

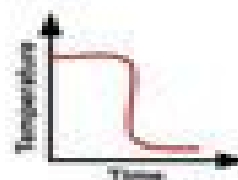
- 22) Which substance underwent an exothermic process? _____
- 23) Which substance underwent an endothermic process? _____
- 24) Which graph below shows the temperature of the steel? _____



Graph A



Graph B



Graph C



Graph D

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M Carnoy



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