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Judd E. Hollander



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web we can write a balanced nuclear equation for this reaction as let s do a practice example identify the product x by writing a balanced nuclear equation $^{137}_{55}\text{Cs} \rightarrow ^{137}_{56}\text{Ba} + x$ because the mass number is the same on both sides of the equation x must have a

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web identify the missing particle in the following nuclear reaction $^1_0\text{n} + ^{235}_{92}\text{U} \rightarrow ^{140}_{54}\text{Xe} + ^{92}_{38}\text{Sr} + 2^1_0\text{n}$ $^{137}_{52}\text{Te} \rightarrow ^{136}_{52}\text{Te} + ^4_2\text{He}$ $^{97}_{38}\text{Sr} \rightarrow ^{96}_{40}\text{Zr} + ^1_0\text{n}$

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