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oxidation $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^-$ reduction $\text{Fe}^{2+} + 2\text{e}^- \rightarrow \text{Fe}$ 24 3 $\text{Ag} \rightarrow \text{Ni}^{3+} + 3\text{e}^-$ oxidation reduction 25 $\text{Cu}^{2+} + \text{Pb} \rightarrow \text{Pb}^{2+} + \text{Cu}$ oxidation reduction 26 O_2

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 $5\text{Co} + \text{I}_2 \rightarrow 5\text{Co}^{2+} + 2\text{I}^-$ $5\text{Co} + \text{I}_2 \rightarrow 5\text{Co}^{2+} + 2\text{I}^-$

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