

3.1 Solution

Given: initial gas volume = $5.667 \times 10^{-3} \text{ m}^3$
 initial gas temperature = 294.3 K
 initial gas pressure = 10.13 bar
 gas doubles in volume to $1.133 \times 10^{-2} \text{ m}^3$
 change in volume = $5.667 \times 10^{-3} \text{ m}^3$

Choose, as the system, the universe or surroundings exclusive of the gas in the cylinder. Then, by the constraints given, including no friction in surroundings:

$$Q = 0$$

and

$$-W = P_a (V_f - V_i) + m_p g \Delta Z$$

where

P_a = atmospheric pressure = 1.013 bar
 $V_f - V_i$ = change in system volume = $-5.667 \times 10^{-3} \text{ m}^3$
 m_p = mass of piston = 225 kg
 g = acceleration due to gravity = 9.81 m/s²
 ΔZ = variation in vertical position of piston
 = -0.3048 m if piston rises
 = 0 if piston moves horizontally
 = +0.3048 m if piston falls

Case 1: Piston rises (note 1 bar = 10^5 N/m^2)

$$W = -(1.013 \times 10^5) (-5.667 \times 10^{-3}) - (225) (9.81) (-0.3048) = +1.250 \text{ kJ}$$

$$\Delta E_{\text{univ}} = Q + W = 1.250 \text{ kJ} = -\Delta E_{\text{gas}} = -N_g C_v (T_f - T_i)$$

$$N_g = \text{moles of gas} = \frac{(10.13 \times 10^5) (5.667 \times 10^{-3})}{(8.314) (294.3)} = 2.346$$

$$C_v = 5 \text{ Btu/lb-mol } ^\circ\text{R} = 20.93 \text{ Btu/mol } ^\circ\text{K}$$

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theory is essential for students who will be asked to fulfill the needs of tomorrow with an increasing demand for talents
which are flexible and adaptable On the other hand the theory is useless unless the student can effectively bridge the gap
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