

Earth Science Reference Tables

PHYSICAL CONSTANTS

Radioactive Decay Data

RADIOACTIVE ISOTOPE	DISINTEGRATION	HALF-LIFE (years)
Carbon-14	$C^{14} \rightarrow N^{14}$	5.7×10^3
Potassium-40	$K^{40} \rightarrow Ar^{40}$ $K^{40} \rightarrow Ca^{40}$	1.3×10^9
Uranium-238	$U^{238} \rightarrow Pb^{206}$	4.5×10^9
Rubidium-87	$Rb^{87} \rightarrow Sr^{87}$	4.9×10^{10}

Specific Heats of Common Materials

MATERIAL	SPECIFIC HEAT (calories/gram • °C)
Water { solid liquid gas	0.5
	1.0
	0.5
Dry air	0.24
Basalt	0.20
Granite	0.19
Iron	0.11
Copper	0.09
Lead	0.03

Properties of Water

Energy gained during melting	80 calories/gram
Energy released during freezing	80 calories/gram
Energy gained during vaporization	540 calories/gram
Energy released during condensation	540 calories/gram
Density at 3.98°C	1.00 gram/milliliter

EQUATIONS

Percent deviation
from accepted value

$$\text{deviation (\%)} = \frac{\text{difference from accepted value}}{\text{accepted value}} \times 100$$

Eccentricity of an ellipse

$$\text{eccentricity} = \frac{\text{distance between foci}}{\text{length of major axis}}$$

Gradient

$$\text{gradient} = \frac{\text{change in field value}}{\text{distance}}$$

Rate of change

$$\text{rate of change} = \frac{\text{change in field value}}{\text{time}}$$

Density of a substance

$$\text{density} = \frac{\text{mass}}{\text{volume}}$$

2001 EDITION

This edition of the Earth Science Reference Tables should be used in the classroom beginning in the 2000-2001 school year. The first examination for which these tables will be used is the January 2001 Regents Examination in Earth Science.


1
2
3
4
5
6
7
8
9
10
11
12
13
14
15
16
17
18
19
20
21
22
23
24
25

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Y. Finkel



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1. Understanding the eBook Use Earth Science Reference Table
 - The Rise of Digital Reading Use Earth Science Reference Table
 - Advantages of eBooks Over Traditional Books
2. Identifying Use Earth Science Reference Table
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Use Earth Science Reference Table
 - User-Friendly Interface
4. Exploring eBook Recommendations from Use Earth Science Reference Table
 - Personalized Recommendations
 - Use Earth Science Reference Table User Reviews and Ratings
 - Use Earth Science Reference Table and Bestseller Lists
5. Accessing Use Earth Science Reference Table Free and Paid eBooks
 - Use Earth Science Reference Table Public Domain eBooks
 - Use Earth Science Reference Table eBook Subscription Services
 - Use Earth Science Reference Table Budget-Friendly Options

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 - Use Earth Science Reference Table Compatibility with Devices
 - Use Earth Science Reference Table Enhanced eBook Features
7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Use Earth Science Reference Table
 - Highlighting and Note-Taking Use Earth Science Reference Table
 - Interactive Elements Use Earth Science Reference Table
8. Staying Engaged with Use Earth Science Reference Table
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Use Earth Science Reference Table
9. Balancing eBooks and Physical Books Use Earth Science Reference Table
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Use Earth Science Reference Table
10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
11. Cultivating a Reading Routine Use Earth Science Reference Table
 - Setting Reading Goals Use Earth Science Reference Table
 - Carving Out Dedicated Reading Time
12. Sourcing Reliable Information of Use Earth Science Reference Table
 - Fact-Checking eBook Content of Use Earth Science Reference Table
 - Distinguishing Credible Sources
13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
 - Integration of Multimedia Elements

- Interactive and Gamified eBooks

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