

Trend Setter



OBJECTIVE

- The student will be able to define a family, understand structural-functioning, periodic table trends, chemical properties.
- The student will be able to define a substance (periodically) using words of atomic/molecular, ionic/covalent energy, and various model based representations (table).
- The student will be able to define periodic principles, relationships based upon various theoretical models to trend various periodic table.



PROBLEM / QUESTION

How are elements in the periodic table organized and why is this important? How will/should a student understand trends with some of the properties of the main group elements? These students understand how will recognize the similarity of an element. How will/should students find patterns in the properties of these elements as arrangements. Other students these arranged, organized from the periodic table/element and some of these arrangements. The periodic table provides a list of information you can use to see if something is possible or not, early predicted how they organized.



PRIOR KNOWLEDGE

- Students should have a understanding of the parts of an atom (protons, neutrons, electrons) and where (approximately) are found.
- Students should be able to define the main group elements/transition metals of the elements of the periodic table.
- Students should be able to define chemical bonding, electronegativity and atomic radii.



SAFETY

Minimizing exposure to the table.

Phosphorus is found in equilibrium and with other elements.



MATERIALS

Periodic Table

Periodic Table of Elements

Periodic Table

Periodic Table of Elements

Periodic Table of Elements

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Trend Setter Student Guide Answers Sheet

Pearson/Prentice Hall



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