# The working cell

Lecture 5: Cellular Metabolism

metalofism - culmination of all the "wo	rk"
that your cell does	+ Heat is produced in most every reaction
-> the use of energy	m Exothermic Teactions -> releases energy
→ importance of enzymes	(exergonic)
-> building or breaking down molecules	tendermic reactions - alosories energence)
movement of fluids or molecules	
	& All transfers of energy loses heat
energy - capacity to do work / couse change	# Energy transfer loses 90% of energy
	ar heat
2 TYPES OF ENERGY	p made of proteins
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* Kinetic energy - energy in motion	perform cell reaction
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-> energy cannot be destroyed nor created	
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-> caused by the friction between atoms & molecule	S Engine Engine substate Engine
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Chemical Energy - where energy comes from	
- molecule in food can be broken	PROPERTIES / FEATURES
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Chris A. Ortiz

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Opportunities and Advancements in Stem Cell Research United States. Congress. House. Committee on Government Reform. Subcommittee on Criminal Justice, Drug Policy, and Human Resources,2002 Updates in Ocular Therapeutics and Surgery, volume III Georgios D. Panos ,Horace Massa,2025-01-27 Ophthalmology is a field of medicine that has grown rapidly during the last few decades Several new and promising treatments have become available for eye conditions including age related macular degeneration glaucoma retinal vein occlusions diabetic macular oedema genetic disorders uveitis and inflammations On the other hand recent technological developments and advanced surgical techniques have drastically changed the daily practice of eye surgeons with new surgical options resulting in safer faster and more precise surgery This Research Topic aims to summarise modern therapeutic and surgical approaches on ocular diseases adding a valuable contribution to our existing knowledge to describe the current state of the art as well as to provide directions for future research

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naming organic compounds 1 1 alkanes a i you shouldn't really have got this one wrong unless you made a careless mistake like putting too many or too few hydrogens on the middle carbon if you got it wrong re read the bit about naming alkanes and then look at these questions again ii

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dec 16 2021 so the complete name is 5 bromo 7 chloro 6 hydroxy 2 2 5 trimethyl 7 octen 4 one 4 it is not difficult to find the parent structure for this compound which is a cyclic alcohol so the last name is cyclopropanol the naming of the substituent with the benzene ring is

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solutions 1 aromatic compound iodobenzene 2 alkane 3 methylhexane 3 ketone 4 alkane alkyl halide 3 chloro 4 ethyl 2 4 dimethyloctane 5 aromatic compound o diethylbenzene or ortho diethylbenzene 6 alkene 3 methylpentene 7 ester 8 aromatic compound p bromotoluene or para bromotoluene 9 alkyne

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jan 23 2023 jim clark truro school in cornwall this page explains how to write the formula for an organic compound given its name and vice versa it covers alkanes cycloalkanes alkenes simple compounds containing halogens alcohols aldehydes and ketones

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feb 21 2017 give the complete iupac name for each of the following compounds a b c give the complete iupac name and the common name for each of the following compounds a b c a 1 ethoxybutane butyl ethyl ether b 1 ethoxypropane ethyl propyl ether c 1 methoxypropane methyl propyl ether

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organic nomenclature quiz this online quiz is intended to give you extra practice in concepts related to introductory organic nomenclature including naming compounds and identifying functional groups from diagrams please note structural isomers are not included in this quiz

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this is a set of practice problems on naming organic compounds the examples cover the nomenclature of alkanes bicyclic compounds alkenes alkynes alcohols alkyl halides aromatic compounds aldehydes and ketones amines ethers and carboxylic acid derivatives such as nitriles esters and amides

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nomenclature of organic compounds chemistry questions with solutions q1 the iupac name of ch 3 cho is ethanal ethanol acetaldehyde acetone answer a the iupac name of ch 3 cho is ethanal q2 the iupac name of ch 3 o c 2 h 5 is methoxy ethane ethoxy ethane methyl ethane

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1 carbon methane 2 carbons ethane 3 carbons propane 4 carbons butane 5 carbons pentane 6 carbons hexane 7 carbons heptane 8 carbons octane 9 carbons nonane 10 carbons decane while many of these names share a greek root with