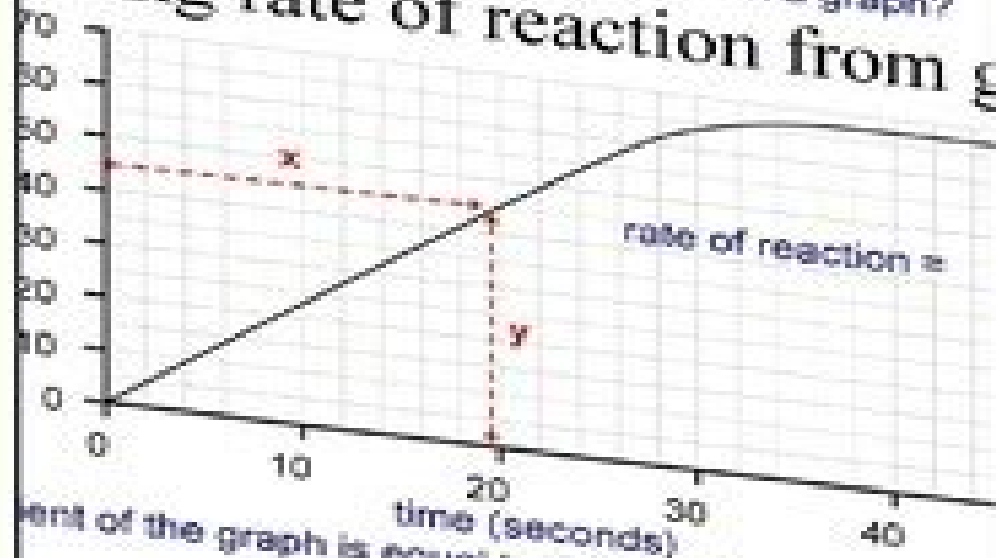


THERMOCHEMISTRY

How can the rate of reaction be calculated from a graph?



What is the rate of reaction at 20 seconds?

rate of reaction = $\frac{45 \text{ cm}^3}{20 \text{ s}}$

REVERSE REACTIONS

As products form, it speeds up the reaction. But as products form, it speeds up the reaction. But as products form, it speeds up the reaction. But as products form, it speeds up the reaction.



How does the rate of reaction change with pressure?

at constant



REACTIONS

Thermochemistry Core Teaching Resources

Camilla Rothe



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