





Chapter - 2

# Acid, Bases And Salts

#### ACIDS:

- These are the substances which have sour taste.
- They turn blue litmus solution red.
- They give H\* ions in aqueous solution.
- The term 'acid' has been derived from the Latin word, acidus, which means sour.

Strong Acids: HCl, H<sub>2</sub>SO<sub>4</sub>, HNO<sub>3</sub>

Weak Acids: CH3COOH, Oxalic acid, Lactic acid

Concentrated Acid: Having more amount of acid + less amount of water

Dilute Acid: Having more amount of water + less amount of acid

#### BASES:

- These are the substances which are bitter in taste and soapy in touch.
- They turn red litmus solution blue.
- They give OH<sup>-</sup> ions in aqueous solution.

Strong Bases: NaOH, KOH, Ca(OH)<sub>2</sub>

Weak Bases: NH<sub>4</sub>OH

Alkalis: These are bases which are soluble in water [NaOH, KOH, Ca(OH)2].

#### SALTS:

These are the compounds formed from reaction of acid and base.





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