

THE PERIODIC TABLE OF ELEMENTS

THE PERIODIC TABLE OF ELEMENTS																			
1																	18		
1	1 H Hydrogen																	2 He Helium	
2	3 Li Lithium	4 Be Beryllium																	10 Ne Neon
3	11 Na Sodium	12 Mg Magnesium																	18 Ar Argon
4	19 K Potassium	20 Ca Calcium	21 Sc Scandium	22 Ti Titanium	23 V Vanadium	24 Cr Chromium	25 Mn Manganese	26 Fe Iron	27 Co Cobalt	28 Ni Nickel	29 Cu Copper	30 Zn Zinc	31 Ga Gallium	32 Ge Germanium	33 As Arsenic	34 Se Selenium	35 Br Bromine	36 Kr Krypton	
5	37 Rb Rubidium	38 Sr Strontium	39 Y Yttrium	40 Zr Zirconium	41 Nb Niobium	42 Mo Molybdenum	43 Tc Technetium	44 Ru Ruthenium	45 Rh Rhodium	46 Pd Palladium	47 Ag Silver	48 Cd Cadmium	49 In Indium	50 Sn Tin	51 Sb Antimony	52 Te Tellurium	53 I Iodine	54 Xe Xenon	
6	55 Cs Cesium	56 Ba Barium	57-71 Lanthanide Series	72 Hf Hafnium	73 Ta Tantalum	74 W Tungsten	75 Re Rhenium	76 Os Osmium	77 Ir Iridium	78 Pt Platinum	79 Au Gold	80 Hg Mercury	81 Tl Thallium	82 Pb Lead	83 Bi Bismuth	84 Po Polonium	85 At Astatine	86 Rn Radon	
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Atomic Number →

1

1.008

← Average Atomic Mass

Symbol →

H

Name →

Hydrogen

Alkali Metals

Alkaline Earth Metals

Transition Metals

Other Metals

Metalloids

Non-metals

Halogens

Noble Gases

Lanthanides

Actinides

What Are The Elements

Alfred Slater West



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lialh oc ch₃ 3 h 4 benzaldehyde c₆h₅cho ch₂ohch₂oh h step 1 toluene oxidation with kmno₄ kmno₄ is a strong oxidising agent which oxidise toluene to benzaldehyde in 1st step

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mnx 7ox4x ex mnx 6ox4x2 basic
m n x 7 o x 4 x e x m n x 6 o x 4 x 2 b a s i c

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