Name:	Class:	Date:	ID: A

Ch 17 Thermochemistry Practice Test

Matching

Match each item with the correct statement below. a. calorimeter enthalpy b. calorie specific heat c. joule heat capacity. quantity of heat needed to raise the temperature of 1 g of water by 1 °C SI unit of energy. quantity of heat needed to change the temperature of 1 g of a substance by 1°C quantity of heat needed to change the temperature of an object by 1°C. 5. device used to measure the heat absorbed or released during a chemical or physical process heat content of a system at constant pressure Match each item with the correct statement below. a. heat of reaction d. heat of fusion b. heat of formation e. heat of solution c. Hess's law of heat summation 7. the enthalpy change for a chemical reaction exactly as it is written 8. the enthalpy change caused by dissolving a substance 9. the energy required to melt a solid at its melting point 10. the change in enthalpy that accompanies the formation of a compound from its elements 11. states that if you add two or more thermochemical equations to give a final equation, you can also add the heats of reaction to give the final heat of reaction

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- 12. What happens to the energy produced by burning gasoline in a car engine?
 - a. The energy is lost as heat in the exhaust.
 - The energy is transformed into work to move the car.
 - c. The energy heats the parts of the engine.
 - d. all of the above
- 13. A piece of metal is heated, then submerged in cool water. Which statement below describes what happens?
 - The temperature of the metal will increase.
 - The temperature of the water will increase.
 - The temperature of the water will decrease.
 - The temperature of the water will increase and the temperature of the metal will decrease.
- 14. How does a calorie compare to a joule?
 - A calorie is smaller than a joule.
 - b. A calorie is larger than a joule.
- c. A calorie is equal to a joule.
- The relationship cannot be determined.

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American Chemical Society

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