



Abridged user's manual

TR1-2013/TR16-2013/TR17-2013/TR2-2013/
TR26-2013/TR27-2013

DECT 6.0 cordless telephone



This Abridged user's manual provides you with basic installation and use instructions.

After reading this manual, you will learn how to:

- ✓ **Install the telephone**
 - 1 Install telephone base & charger
 - 2 Install handset battery
- ✓ **Program common settings**
 - 3 Use built-in answering system or voicemail?
 - 4 Set date and time
 - 5 Set handset ringer volume
 - 6 Set quiet mode
- ✓ **Use common operations**
 - 7 Directory
 - 8 Speed dial
 - 9 Caller ID
 - 10 Answering system

We recommend you refer to the full set of installation and operation instructions at www.vtechphones.com. For customer service, visit www.vtechphones.com or call 1 (800) 595-9511.

Before using this VTech product, please read **Important safety instructions** on page 10 of this user's manual.

Vtech Phone Instructions Manual

Clifford Lane



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web the relative electronegativities of the bonded atoms is h c cl and so the bond moments all point toward the cl end of the
 molecule and sum to yield a considerable dipole moment the molecules are relatively polar for molecules of high symmetry
 such as bf₃ trigonal planar ch₄ tetrahedral pf₅ trigonal bipyramidal and sf₆

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web explain the difference between a nonpolar covalent bond a polar covalent bond and an ionic bond answer nonpolar
 covalent electronegativity difference is less than 0 4 nonmetal nonmetal close together on the periodic table

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web apr 1 2021 updated on april 01 2021 chemical bonds may be classified as being either polar or nonpolar the difference
 is how the electrons in the bond are arranged key takeaways what is a polar bond in chemistry a polar bond is a type of
 covalent bond in which the electrons forming the bond are unequally distributed