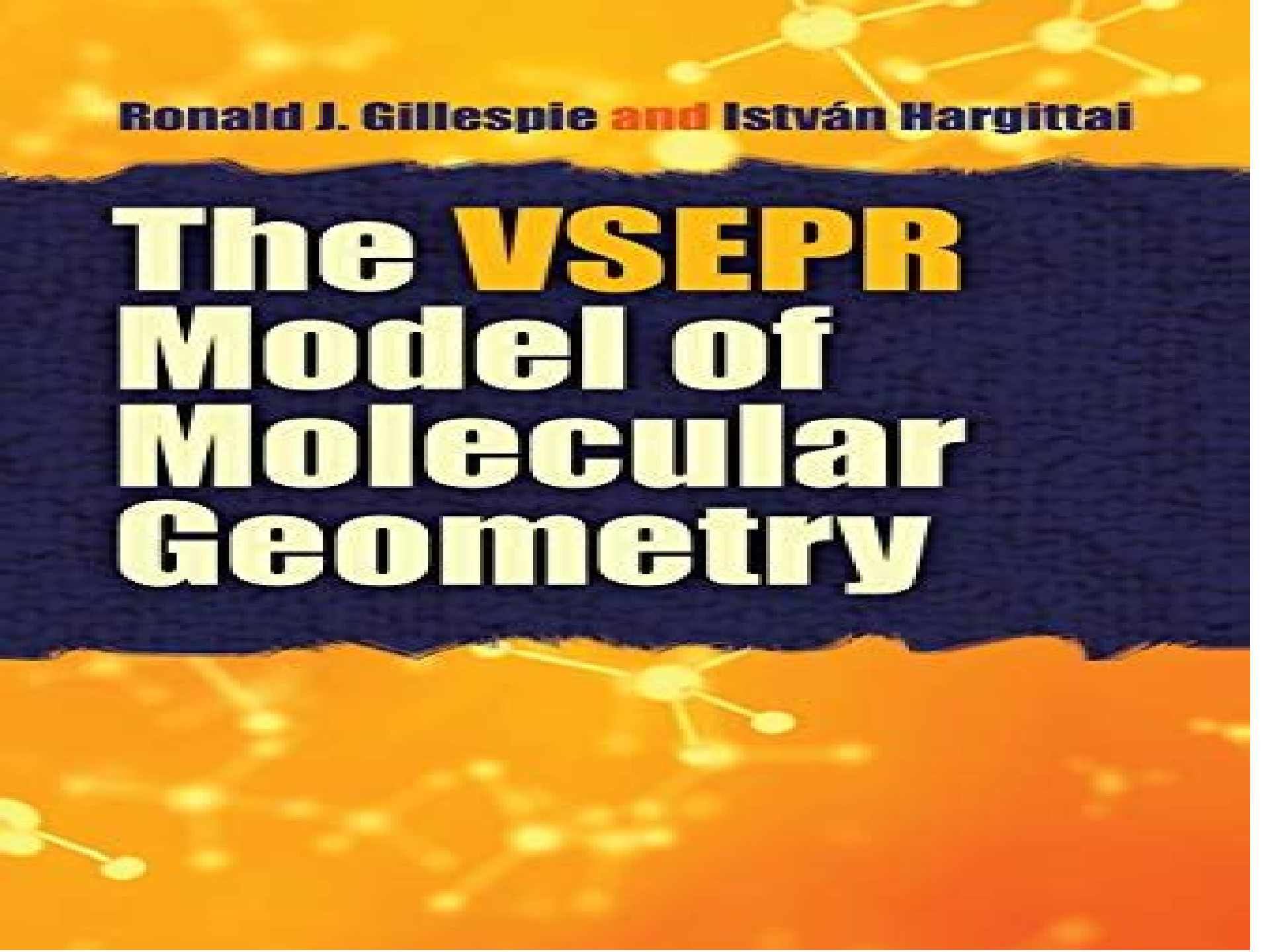


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Ronald J. Gillespie and István Hargittai

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anyone in the world The chapters feature such famous contributors to world science as Francis Crick Richard L Garwin Ronald J Gillespie Avram Hershko George Klein Paul Lauterbur and Peter Mansfield Peter D Lax Paul Nurse Yuval Ne eman George A Olah Guy Ourisson Michael Polanyi Andrei D Sakharov Albert Szent Gy rgyi Edward Teller James D Watson and Eugene P Wigner The areas covered include chemistry molecular biology physics materials science and mathematics On the basis of Hargittai s mosaic of his personal and scientific life I could compose two further patterns One would be the history of the twentieth century and the other the science history of the same time period From the Foreword by the late philosopher Agnes Heller Goethe Medalist Wallenberg Medalist and Hannah Arendt Prize laureate *The Cumulative Book Index* ,1992 A world list of books in the English language **Subject Guide to Books in Print** ,1993 Molecular Geometry Alison Rodger,Mark Rodger,2014-05-16 Molecular Geometry discusses topics relevant to the arrangement of atoms The book is comprised of seven chapters that tackle several areas of molecular geometry Chapter 1 reviews the definition and determination of molecular geometry while Chapter 2 discusses the unified view of stereochemistry and stereochemical changes Chapter 3 covers the geometry of molecules of second row atoms and Chapter 4 deals with the main group elements beyond the second row The book also talks about the complexes of transition metals and f block elements and then covers the organometallic compounds and transition metal clusters The last chapter tackles the consequences of small local variations in geometry The text will be of great use to chemists who primarily deal with the properties of molecules and atoms

Models of Molecular Shapes/VSEPR Theory and Orbital Hybridization James M. Postma,Julian Roberts,J. Leland Hollenberg,2000-01-15 **Molecules and Models** Arne Haaland,2008-03-06 This book provides a systematic description of the molecular structures and bonding in simple compounds of the main group elements with particular emphasis on bond distances bond energies and coordination geometries The description includes the structures of hydrogen halogen and methyl derivatives of the elements in each group some of these molecules are ionic some polar covalent The survey of molecules whose structures conform to well established trends is followed by representative examples of molecules that do not conform We also describe electron donor acceptor and hydrogen bonded complexes Chemists use models to systematize our knowledge to memorize information and to predict the structures of compounds that have not yet been studied The book provides a lucid discussion of a number of models such as the Lewis electron pair bond and the VSEPR models the spherical and polarizable ion models and molecular orbital calculations and it outlines the successes and failures of each

Sacrificios Virginia Gil de Hermoso,1962 Molecular Geometry Ronald James Gillespie,1972-01-01 Molecular Modelling for Beginners Alan Hinchliffe,2005-12-17 Presenting a concise basic introduction to modelling and computational chemistry this text includes relevant introductory material to ensure greater accessibility to the subject Provides a comprehensive introduction to this evolving and developing field Focuses on MM MC and MD with an entire chapter devoted to QSAR and Discovery Chemistry Includes many real chemical applications combined with worked problems and solutions

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Shape in Chemistry Paul G. Mezey, 1993 The Molecular Geometries of Coordination Compounds in the Vapour Phase Magdolna Hargittai, István Hargittai, 1977 Models, Mysteries, and Magic of Molecules Jan C. A. Boeyens, J.F. Ogilvie, 2007-11-03 Molecules have for a long time been of central importance in chemistry as the basis on which all new products and materials have been designed developed and interpreted Since the discovery and characterization of active biomolecules biology has also been transformed into a molecular science With the new developments of molecular devices single molecule spectroscopy time resolved x ray diffraction and the study of mass selected clusters in molecular beams materials science and electronics may move in the same direction The understanding of molecules and the dynamics of their transition between isolated and assembled states rests on three pillars structure activity and function Enormous progress has been made in the experimental study of molecules by diffraction and spectroscopic analysis directed at all three of the basic aspects In the process molecular scientists have developed efficient working models

in terms of which to design and interpret their experiments A vital feature of such models is an understanding of intra and intermolecular cohesion and assembly or chemical bonding One challenging aspect is that the working models of chemistry are notoriously difficult to reduce to the fundamental theories of the physical sciences The Indaba 5 meeting held in South Africa during August 2006 was aimed at the progress that is being made to achieve first principle understanding of molecular science structure and dynamics that includes bonding activity and function An interdisciplinary international team of experts came together to document and to probe various aspects of these fundamental questions and their startling conclusions confirm the need for a fresh look at the physical sciences with a view to better understand the mysteries and magic of molecules This book explores the common ground to guide chemists biologists crystallographers spectroscopists and theorists into a deeper recognition of their individual relevance towards painting a holistic picture of scientific endeavour This effort to stimulate interest in multidisciplinary research is rare if not unique

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