

GRAPHING TRENDS OF THE PERIODIC TABLE

CHEMISTRY

NAME: _____

DATE: _____

PERIOD: _____

Graphing Periodic Table Trends

Directions: Using the provided graph paper, graph each of the elemental properties using a line graph. Place the property on the Y-axis and the Atomic number on the X-axis. Draw a dark, solid, vertical line on each graph where each row on the periodic table ends. (i.e. after atomic number 10). Use the graphs to analyze the trends in these properties and to answer the questions on the back.

Atomic Radius (pm)							
Lithium - 3	152	Sodium - 11	190	Potassium - 19	243	Rubidium - 27	265
Beryllium - 4	112	Magnesium - 12	145	Calcium - 20	194	Strontium - 28	219
Boron - 5	87	Aluminum - 13	118	Gallium - 31	136	Indium - 49	156
Carbon - 6	67	Silicon - 14	111	Germanium - 52	125	Tin - 50	145
Nitrogen - 7	56	Phosphorus - 15	98	Arsenic - 33	114	Antimony - 51	133
Oxygen - 8	48	Sulfur - 16	88	Selenium - 34	103	Tellurium - 52	123
Fluorine - 9	42	Chlorine - 17	79	Bromine - 35	94	Iodine - 53	115

Ionization Energy (kJ/mol)							
Lithium - 3	520	Sodium - 11	496	Potassium - 19	419	Rubidium - 27	403
Beryllium - 4	900	Magnesium - 12	738	Calcium - 20	590	Strontium - 28	550

Atomic Radius



Trends In The Periodic Table Graphing

American Chemical Society



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Computer Based Projects for a Chemistry Curriculum Thomas J. Manning,Aurora P. Gramatges,2013-04-04 This e book is a collection of exercises designed for students studying chemistry courses at a high school or undergraduate level The e book contains 24 chapters each containing various activities employing applications such as MS excel spreadsheets and Spartan computational modeling Each project is explained in a simple easy to understand manner The content within this book is suitable as a guide for both teachers and students and each chapter is supplemented with practice guidelines and exercises Computer Based Projects for a Chemistry Curriculum therefore serves to bring computer based learning a much needed addition in line with modern educational trends to the chemistry classroom *Chemistry in the Community* American Chemical Society,2006-01-31 This laboratory based text centres itself around decision making activities where students apply their chemistry knowledge to realistic situations This fifth edition includes more photographs new drawings and new design

Basic Concepts of Chemistry Leo J. Malone,Theodore O. Dolter,2011-12-27 The 9th edition of Malone s Basic Concepts of Chemistry provides many new and advanced features that continue to address general chemistry topics with an emphasis on outcomes assessment New and advanced features include an objectives grid at the end of each chapter which ties the objectives to examples within the sections assessment exercises at the end each section and relevant chapter problems at the end of each chapter Every concept in the text is clearly illustrated with one or more step by step examples Making it Real essays have been updated to present timely and engaging real world applications emphasizing the relevance of the material they are learning This edition continues the end of chapter Student Workshop activities to cater to the many different learning styles and to engage users in the practical aspect of the material discussed in the chapter WileyPLUS sold separately from text **Science Units for Grades 9-12** Randy L. Bell,Joe Garofalo,2005 Sample topics include cell division virtual dissection earthquake modeling the Doppler Effect and more

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commercial products numerous photographs of reactions in chapter practice exercises following worked examples

ChemCom American Chemical Society,1993 *Catalog of Instructional Films (16mm, Super 8, and 8mm) for College Chemistry* Advisory Council on College Chemistry,1969 *Holt Chemistry* Ralph Thomas Myers,2004 Prentice Hall Chemistry ,2000 **Physical Science** Saddleback Educational Publishing,2013-09-01 Physical science includes any of the sciences such as physics chemistry astronomy and geology that analyze the nature and properties of energy and nonliving matter and seeks to expand the physical and mathematical understanding of the universe The reproducible activity pages supplement physical science textbooks with stand alone or coordinate one page lessons Sample activities include Acceleration Changes of State Density Electric Charges Fixed and Moveable Pulleys Heat Engines Hydrocarbons Solutions and More Chemistry Martin Stuart Silberberg,1996 This general chemistry text centres on the theme that observable change in chemical systems is the result of molecular change The aims of this edition are to enable students to perceive matter and change at the molecular level and to help build student confidence in their ability to solve chemical problems as they discover the relevance of chemistry to their lives **Instructional Media and the New Technologies of Instruction** Robert Heinich,Michael Molenda,James D. Russell,1993 Grade level 1 2 3 4 5 6 7 8 9 10 11 12 e i s t **Laboratory Manual** Jo A. Beran,1990

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