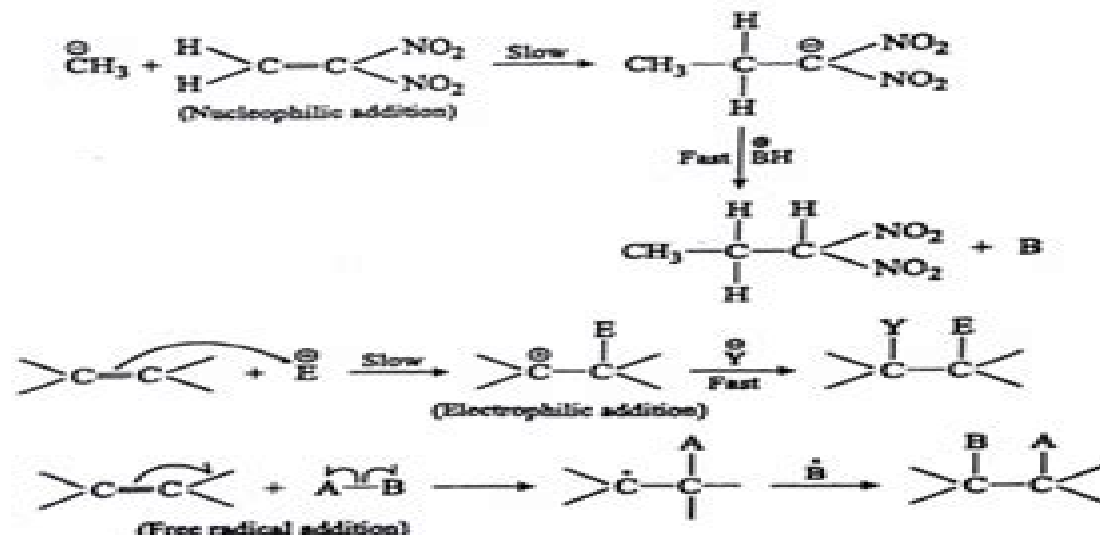


(2) Addition Reactions

A reaction in which two molecules combine to give a single molecule of product is known as **addition reaction**. Addition reactions involve an increase in the number of groups attached to the substrate, and therefore a decrease in the degree of unsaturation of the substrate is an addition reaction. Most commonly, an addition involves the gain of two atoms or groups (one electrophile and one nucleophile, or both free radicals) at each end of the bond or ends of a system. As with substitution reaction, here also the attacking species may be a nucleophile, electrophile or free radical.



(3) Elimination Reactions

The reactions involving removal of atoms or groups from a molecule without their replacement by other atoms or groups are known as **elimination reactions**.



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