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praekingen one poase en
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Twée Schelven Hooi Opmerkingen Over Pozie En Politiek

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In het licht van de kritiek Susanne Janssen, 1994 Tevens verschenen als proefschrift Aandacht voor de omgang van de dag en weekbladkritiek met het literaire aanbod Er wordt ingegaan op de keuzen en uitspraken van critici in samenhang met de sociale context waarbinnen ze tot stand komen In drie case studies worden de literair kritische receptie de critici en de rol van achtereenvolgens Rutger Kopland Judicus Verstegen en Jacq Firmin Vogelaar besproken *Twee schelven hooi* Aad Nuis, 1968 **Het masker van de redelijkheid** K. L. Poll, 1969 Gebundelde kritieken en essays *Volmaakt onaf* Stefaan Evenepoel, 2000 *The Year's Work in Modern Language Studies*, 1931 **The British Library general catalogue of printed books to 1975** British Library, 1984 Subject Index of Modern Books Acquired British Library, 1971 *Vrij Nederland*, 2007 *Literair lustrum: 1966-1971* Kees Fens, H. U. Jessurun d'Oliveira, J. J. Oversteegen, 1967 Verzameld werk Wilfred Smit, 1983 **Hollands maandblad**, 1968 **Library of Congress Catalog** Library of Congress, 1965 A cumulative list of works represented by Library of Congress printed cards **Kritisch lexicon van de Nederlandstalige literatuur na 1945** Ad Zuiderent, Hugo Brems, T. van Deel, 1980 **De gids**, 1968 **Books Abroad**, 1969 *Schrijvers prentenboek*, 2003 **Subject Catalog** Library of Congress, 1965 **Levende Talen**, 1968 *Venus in minirok* Piet Calis, 2010 Overzicht van de manier waarop de seksualiteit beschreven wordt in de Nederlandstalige literatuur vanaf 1945 **Raam**, 1974

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for the nitration 4.01 ml of nitric acid 4.00 ml of sulfuric acid and 4.56 ml bromobenzene were used to complete the reaction with these measurements bromobenzene was found to be the limiting reagent since only 4.56 ml of bromobenzene was used the amount of product predicted to be recovered was 8.7533 g

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the aim of the experiment is to determine and study electrophilic aromatic substitution reaction especially nitration of methyl benzoate to give methyl m nitrobenzoate thus determine the weight melting point and percentage yield of the pure methyl m nitrobenzoate abstract

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in this experiment we will carry out the nitration of bromobenzene bromine substituted on an aromatic ring is an ortho para directing group it will direct an incoming electrophile to the ortho and para positions it is also deactivating meaning that its reactions will be slower than the identical reactions of benzene

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1 go to the restroom and wash the affected area with lots of soap and water the soap in the lab is not suitable for this purpose 2 return to the lab and apply mineral oil to the affected area 3 the summary to this warning is not to touch any part of your body while performing this experiment if you must leave the lab for any reason 1

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in the nitration of bromobenzene a mixture of sulfuric acid and nitric acid react with one another nitric acid the weaker of the two acids will serve as base to form nitronium ion bromobenzene will attack the nitronium ion by nucleophilic attack

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then actual is your actual findings yield actual theoretical $\times 100$ what are the possible side reactions of nitration of bromobenzene dinitration can occur at ortho para or at both ortho positions occurs in overheating meta product happens very rarely because bromine is an ortho para director

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the bromine substituent in bromobenzene is an ortho para director the para position is favored due to steric effects in the reaction that takes place in this lab a nitro group is substituted onto bromobenzene at the para position to form 1-bromo-4-nitrobenzene

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over 5 minutes slowly pipet this bromobenzene into the icy cold nitration mixture HNO_3 3 mL H_2SO_4 in the 125 mL flask and stir continue to stir the reaction mixture in the ice bath for 30 minutes do not heat in a new 250 mL erlenmeyer

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on the right the arenium cation has a nitro set with incredibly small 0.6 angstrom bond distances this will result in an optimization failure as the atoms are even close for the programme to accomodate their electrons demonstration of creating the p arenium cation by to nitration of bromobenzene

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the purpose of this lab was to observe where substituents are added on a benzene ring by way of electrophilic aromatic substitution the electrophilic source in the nitration of bromobenzene comes from a nitronium ion NO_2^+ which acts as the electrophilic species

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nitration of bromobenzene is an electrophilic aromatic substitution reaction the electrophile will be formed from 4.0ml of nitric acid and 4.0ml of sulfuric acid and will then react with 4.5ml of bromobenzene this will produce o bromonitrobenzene and p bromonitrobenzene

experiment 11 electrophilic aromatic substitution nitration - Feb 10 2022

experimental spectra reaction a fluorobenzene stock fluorobenzene 1 h 1 h 19 f 19 f 13 c nmr fid for reference only sample 1 h 1 h 19 f 19 f 13 c nmr eas nitration of fluorobenzene crude product not available for submission for credit

nitration of bromobenzene chem 154 gwu studocu - Mar 26 2023

lab writeup the purpose of this lab was to observe where substituents are added on benzene ring way of electrophilic aromatic substitution the electrophilic

5 nitration lab docx lab 5 nitration of bromobenzene - Aug 19 2022

chem 2261 lab 5 nitration of bromobenzene introduction an electrophilic aromatic substitution reaction is an important process that allows for intermediate chemical structures to be used as precursors to pharmaceuticals and other industrially produced chemicals by adding different functional groups into an aromatic ring

[lab report nitration of bromomobenzene ipl org](#) - Feb 22 2023

lab report 10 nitration of bromobenzene raekwon filmore cm 244 section 40 march 27 2018 introduction for this experiment

nitration of bromobenzene was the focus of the lab the benzene is an aromatic compound and when it reacts with wither a mixture of sulfuric acid or nitric acid creates what is known as a nitro group