

Lab Report: Thermal Energy Transfer

Experiment 1 Part 1: When objects are in direct contact heat is transferred from the object with higher temperature to the object with lower temperature. This experiment aims to determine the amount of heat transferred.

Experiment 1 Part 2: In this experiment, we will determine the amount of heat transferred from a hot object to a cold object through conduction.

Material	Initial Temp (°C)	Final Temp (°C)	Heat Transferred (J)	Notes
Water	20.0	25.0	100 J	10 g
Water	20.0	25.0	100 J	10 g
Water (Total)	20.0	25.0	200 J	20 g
Water (Total)	20.0	25.0	200 J	20 g
Hot Water	20.0	25.0	100 J	10 g
Cold Water	20.0	25.0	100 J	10 g

Calculate the amount of heat transferred from the hot water to the cold water using the formula: $Q = mc\Delta T$.

Material	Initial Temp (°C)	Final Temp (°C)	Heat Transferred (J)	Notes
Water	20.0	25.0	100 J	10 g
Water	20.0	25.0	100 J	10 g
Water (Total)	20.0	25.0	200 J	20 g
Water (Total)	20.0	25.0	200 J	20 g
Hot Water	20.0	25.0	100 J	10 g
Cold Water	20.0	25.0	100 J	10 g

Material	Initial Temp (°C)	Final Temp (°C)
Water	20.0	25.0
Water	20.0	25.0

Thermal Effects In Semioconductors Lab Manual

Jin-Ying Zhang



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