

- A) acid or a base.
- B) neutral molecule or an ion.
- C) polar or a nonpolar molecule.
- D) standard or a nonstandard monomer in proteins.
- E) transparent or a light-absorbing compound.

10. Amino acids**Pages: 79–80****Difficulty: 2****Ans: D**

Titration of valine by a strong base, for example NaOH, reveals two pK 's. The titration reaction occurring at pK_2 ($pK_2 = 9.62$) is:

- | | | |
|-------------------------------------|---------------|--|
| A) $-\text{COOH} + \text{OH}^-$ | $\rightarrow$ | $-\text{COO}^- + \text{H}_2\text{O}$. |
| B) $-\text{COOH} + -\text{NH}_2$ | $\rightarrow$ | $-\text{COO}^- + -\text{NH}_2^+$. |
| C) $-\text{COO}^- + -\text{NH}_2^+$ | $\rightarrow$ | $-\text{COOH} + -\text{NH}_2$. |
| D) $-\text{NH}_3^+ + \text{OH}^-$ | $\rightarrow$ | $-\text{NH}_2 + \text{H}_2\text{O}$. |
| E) $-\text{NH}_2 + \text{OH}^-$ | $\rightarrow$ | $-\text{NH}^- + \text{H}_2\text{O}$. |

11. Amino acids**Pages: 79–80****Difficulty: 1****Ans: C**

In a highly basic solution, $\text{pH} = 13$, the dominant form of glycine is:

- A) $\text{NH}_2-\text{CH}_2-\text{COOH}$.
- B) $\text{NH}_2-\text{CH}_2-\text{COO}^-$.
- C) $\text{NH}_2-\text{CH}_2^+-\text{COO}^-$.
- D) $\text{NH}_3^+-\text{CH}_2-\text{COOH}$.
- E) $\text{NH}_3^+-\text{CH}_2-\text{COO}^-$.

12. Amino acids**Pages: 80–81****Difficulty: 2****Ans: B**

For amino acids with neutral R groups, at any pH below the pI of the amino acid, the population of amino acids in solution will have:

- A) a net negative charge.
- B) a net positive charge.
- C) no charged groups.
- D) no net charge.
- E) positive and negative charges in equal concentration.

13. Amino acids**Pages: 80–81****Difficulty: 2****Ans: A**

At $\text{pH} 7.0$, converting a glutamic acid to γ -carboxyglutamate, will have what effect on the overall charge of the protein containing it?

- A) It will become more negative.
- B) It will become more positive.
- C) It will stay the same.
- D) There is not enough information to answer the question.
- E) The answer depends on the salt concentration.

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generate writing idea or simply encourage speaking