

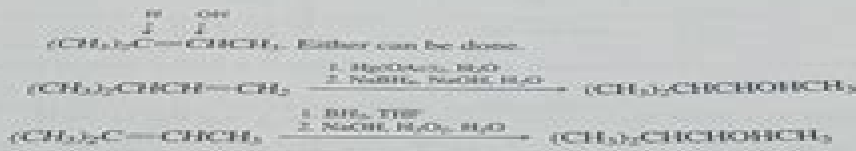
All products are formed in varying amounts.



Note the anti-Markovnikov regiochemistry for hydroboration.

43. A brief analysis of possible choices is provided for each problem.

(a) Need Markovnikov hydration of $(CH_3)_2CHCH=CH_2$, or anti-Markovnikov hydration of



(b) Need addition to propene of " Cl^+ " and " $(CH_3)_2CHO^-$." Cl_2 is the source of " Cl^+ ", and $(CH_3)_2CHOH$ provides the nucleophile.



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K. Peter C. Vollhardt, Neil E. Schore



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