

Matching

Match each item with the correct statement below.

- | | |
|---------------------|----------------|
| A. solvation | E. electrolyte |
| B. weak electrolyte | F. colloid |
| C. aqueous solution | G. surfactant |
| D. solvent | |

- _____ 1. interferes with hydrogen bonding between water molecules
- _____ 2. dissolving medium
- _____ 3. homogeneous mixture of water and dissolved substances
- _____ 4. Solute ions or molecules are surrounded by solvent molecules.
- _____ 5. compound that will conduct current in the liquid state or in aqueous solution
- _____ 6. compound that ionizes incompletely in aqueous solution
- _____ 7. mixture in which particle size averages between 1 nm and 1000 nm

Match each item with the correct statement below.

- | | |
|----------------------|-------------------|
| A. dispersed phase | E. Tyndall effect |
| B. surface tension | F. suspension |
| C. Brownian motion | G. solute |
| D. dispersion medium | H. emulsion |

- _____ 8. inward force tending to minimize surface area of a liquid
- _____ 9. dissolved particle
- _____ 10. mixture in which particle size averages greater than 1000 nm in diameter
- _____ 11. Colloidal particles spread throughout a suspension.
- _____ 12. phenomenon observed when beam of light passes through a colloid
- _____ 13. chaotic movement of colloidal particles
- _____ 14. colloid of a liquid in a liquid

Multiple Choice

Identify the choice that best completes the statement or answers the question.

- _____ 15. How does the surface tension of water compare with the surface tensions of most other liquids?
 - A. It is lower.
 - B. It is about the same.
 - C. It is higher.
 - D. It is higher only when a surfactant is added.
- _____ 16. What causes water's low vapor pressure?

A. dispersion forces	C. hydrogen bonding
B. covalent bonding	D. ionic attractions
- _____ 17. What is the shape of the water molecule?

A. linear	B. tetrahedral	C. trigonal planar	D. bent
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- _____ 18. Which of the following is primarily responsible for holding water molecules together in the liquid state?

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Pandey S.N. & Sinha B.K.



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