

Balancing Word and Skeleton Equations

Write and balance the skeleton equation for each of the following reactions.

- a** calcium carbide + oxygen $\rightarrow$ calcium + carbon dioxide
- b** chromium + tin (IV) chloride $\rightarrow$ chromium (III) chloride + tin
- c** potassium hydroxide + hydrogen bromide $\rightarrow$ potassium bromide + water
- d** aluminum oxide $\rightarrow$ aluminum + oxygen gas
- e** propane + oxygen $\rightarrow$ carbon dioxide + water
- f** barium carbonate $\rightarrow$ barium oxide + carbon dioxide
- g** iron + sulfur $\rightarrow$ iron (II) sulfide
- h** aluminum + copper (II) chloride $\rightarrow$ aluminum chloride + copper
- i** silver nitrate + copper $\rightarrow$ silver + copper nitrate
- j** hydrogen + chlorine $\rightarrow$ hydrogen chloride

Word And Skeleton Equations Answers

**Herman J. Bruckman, Allan
Cruickshanks**



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