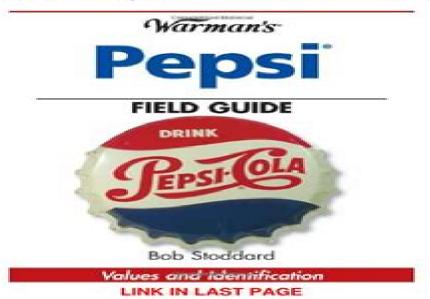
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Joe Soucy

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web mar 4 2014 contains water the hand warmer is struck in a manner that ruptures the inner pouch releasing the ionic salt into the water of the outer pouch the salt dissolves and the water warms this type of hand warmer tends to produce a more vigorous heat than the dry powder type of hand warmer but does not produce heat for quite as long

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web feb 2 2017 this is a video outlining a flinn lab on designing a hand warmer it was created by alex brinley charis conwell and siena joy for our ap chemistry class

explore the chemistry within hand warmers science project - Sep 23 2022

web hand warmers provide a unique and fun way to study the chemistry of crystal formation and heat generation by pressing a button in a pouch which contains a supercooled solution you start a rapid exothermic heat producing crystallization in this science fair project you will determine how the starting temperature affects hand warmer chemistry

ap chemistry hand warmer lab youtube - Oct 25 2022

web feb 1 2017 hannah boster seth lewis and william makinen s recorded powerpoint presentation summarizing our work on the flinn handwarmer lab created for mr kerns ap

12 the hand warmer design challenge studocu - Nov 25 2022

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ap chemistry lab 10 hand warmer design challenge big idea 5 - Oct 05 2023

web lab investigation in today s experiment you will design and execute an experimental procedure to determine which of three ionic compounds is most suitable for use in a hand warmer the three compounds you will use is determined by the group number you are assigned the ideal hand warmer