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Chapter 4, Section 4: Reaction types – Acid-Base reactions

EX 12: What volume of 0.100 M HCl solution is needed to neutralize 25.0 mL of 0.350 M NaOH?

Then replace HCl with H₂SO₄. Then replace HCl with H₂SO₄. Then replace HCl with H₂SO₄. Then replace HCl with H₂SO₄.

EX 13: In an experiment, 28 mL of 0.250 M HNO, and 53 mL of 0.320 M KOH are mixed. Calculate the amount of water formed in the reaction and calculate the concentration of H' or OH in excess after the reaction goes to completion.

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