# Thermochemical Equations

### Study Guide

### **Key Concepts**

- A thermochemical equation includes the physical states of the reactants and products and specifies the change in enthalpy.
- The molar enthalpy (heat) of vaporization, ΔH<sub>vap</sub>, is the amount of energy required to evaporate one mole of a liquid.
- The molar enthalpy (heat) of fusion, ΔH<sub>fus</sub>, is the amount of energy needed to melt one mole of a solid.







## **Thermochemical Equations Study Guide Key**

**RM Cervero** 

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