

MECHANICS OF FLUIDS

1. Determine the mass and the weight of the air contained in a room whose dimensions are 6 m x 6 m x 8 m. Assume the density of the air is 1.16 kg/m^3 .

1. Determined the mass and the weight of the air contained in a room whose dimensions are 6m x 6m x 8m. Assume the density of the air is 1.16 kg/m^3 .

Given :

$$\rho = 1.16 \text{ kg/m}^3 \longrightarrow \text{Density of air}$$

$$g = 9.81 \text{ m/s}^2 \longrightarrow \text{Assume gravity acceleration}$$

$$6\text{m} \times 6\text{m} \times 8\text{m} \longrightarrow \text{Dimension of room}$$

Solutions:

Volume of room:

$$V = 6\text{m} \times 6\text{m} \times 8\text{m} = 288 \text{ m}^3$$

Mass is defined by volume of room & air :

$$m = \rho \cdot V = \left(1.16 \frac{\text{kg}}{\text{m}^3}\right) (288 \text{ m}^3) = \boxed{334.08 \text{ kg}}$$

Weight of object :

$$G = m \cdot g = (334.08 \text{ kg}) (9.807 \frac{\text{m}}{\text{s}^2}) \left(\frac{1 \text{ N}}{1 \text{ kg} \cdot \frac{\text{m}}{\text{s}^2}}\right) = \boxed{3277.33 \text{ N}}$$

Therefore :

$$m = 334.08 \text{ kg}$$

$$G = 3277.33 \text{ N}$$

Armaline D. Quillas ~~Armaline D. Quillas~~ P.

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Solver Max Fogiel, Ralph W. Pike, Research and Education Association, 1993 REA's Thermodynamics Problem Solver Each Problem Solver is an insightful and essential study and solution guide chock full of clear concise problem solving gems Answers to all of your questions can be found in one convenient source from one of the most trusted names in reference solution guides More useful more practical and more informative these study aids are the best review books and textbook companions available They're perfect for undergraduate and graduate studies This highly useful reference provides thorough coverage of pressure work and heat energy entropy first and second laws ideal gas processes vapor refrigeration cycles mixtures and solutions For students in engineering physics and chemistry

Thermodynamics William Z. Black, James G. Hartley, 1996 The third edition of Thermodynamics provides an easily understandable presentation of classical thermodynamics that builds on the student's background of energy concepts first learned in physics and chemistry The material is organized in a logical progression from the conservation of mass the conservation of energy and the second law The engineering perspective is retained and a variety of familiar examples are used so that the student can appreciate how thermodynamics affects a broad range of subjects The authors continue to emphasize a systematic approach to problem solving and that approach is used in all example problems in the text This problem solving method provides not only a reasonable way to approach the task of solving thermodynamics problems but it also will serve the student in other engineering and science disciplines Each example is worked in detail and particular attention has been given to the proper use of units and unit conversions in the solutions Detailed explanations accompany the simplifications when the general equations are reduced to the forms that apply to special cases so that the student will gain a better understanding of the conservation principles as well as greater awareness of these powerful analytical tools Examples address the questions of which form of the conservation laws should be used and why certain assumptions can be applied to simplify the solutions Believing that second law analysis should play a major role in the analysis of engineering problems the authors provide extensive coverage of the second law of thermodynamics The development of the second law is similar to that used for the introduction of the conservation of mass and energy The results of the second law are carried over into subsequent chapters where they are applied to thermodynamic systems such as power and refrigeration cycles as well as air conditioning processes

Problems And Solutions On Thermodynamics And Statistical Mechanics Yung-kuo Lim, 1990-02-01 The material for these volumes has been selected from the past twenty years examination questions for graduate students at University of California at Berkeley Columbia University the University of Chicago MIT State University of New York at Buffalo Princeton University and University of Wisconsin

Problems in Chemical Thermodynamics with Solutions Maka Aleksishvili, Shota Sidamonidze, 2002 The methods of chemical thermodynamics are effectively used in many fields of science and technology Mastering these methods and their use in practice requires profound comprehension of the theoretical questions and acquisition of certain calculating skills This book is useful to undergraduate and graduate students in chemistry as well as

chemical thermal and refrigerating technology it will also benefit specialists in all other fields who are interested in using these powerful methods in their practical activities

Modern Engineering Thermodynamics - Textbook with Tables Booklet Robert T. Balmer,2011-01-03 Modern Engineering Thermodynamics Textbook with Tables Booklet offers a problem solving approach to basic and applied engineering thermodynamics with historical vignettes critical thinking boxes and case studies throughout to help relate abstract concepts to actual engineering applications It also contains applications to modern engineering issues This textbook is designed for use in a standard two semester engineering thermodynamics course sequence with the goal of helping students develop engineering problem solving skills through the use of structured problem solving techniques The first half of the text contains material suitable for a basic Thermodynamics course taken by engineers from all majors The second half of the text is suitable for an Applied Thermodynamics course in mechanical engineering programs The Second Law of Thermodynamics is introduced through a basic entropy concept providing students a more intuitive understanding of this key course topic Property Values are discussed before the First Law of Thermodynamics to ensure students have a firm understanding of property data before using them Over 200 worked examples and more than 1 300 end of chapter problems provide an extensive opportunity to practice solving problems For greater instructor flexibility at exam time thermodynamic tables are provided in a separate accompanying booklet University students in mechanical chemical and general engineering taking a thermodynamics course will find this book extremely helpful Provides the reader with clear presentations of the fundamental principles of basic and applied engineering thermodynamics Helps students develop engineering problem solving skills through the use of structured problem solving techniques Introduces the Second Law of Thermodynamics through a basic entropy concept providing students a more intuitive understanding of this key course topic Covers Property Values before the First Law of Thermodynamics to ensure students have a firm understanding of property data before using them Over 200 worked examples and more than 1 300 end of chapter problems offer students extensive opportunity to practice solving problems Historical Vignettes Critical Thinking boxes and Case Studies throughout the book help relate abstract concepts to actual engineering applications For greater instructor flexibility at exam time thermodynamic tables are provided in a separate accompanying booklet

Heat Transfer Kubie Jorge,Tariq Muneer,Grassie Thomas,2012-08-06 A core task of engineers is to analyse energy related problems The analytical treatment is usually based on principles of thermodynamics fluid mechanics and heat transfer but is increasingly being handled computationally This unique resource presents a practical textbook written for both undergraduates and professionals with a series of over 60 computer workbooks on accompanying downloadable resources The book emphasizes how complex problems can be deconstructed into a series of simple steps All thermophysical property computations are illustrated using diagrams within text and on the downloadable resources

Problems and Solutions in Engineering Thermodynamics S C. Singhal,2009-02-01 *Problems in Thermodynamics and Statistical Physics* Peter T. Landsberg,2014-07-16 Well

respected and widely used this volume presents problems and full solutions related to a wide range of topics in thermodynamics statistical physics and statistical mechanics The text is intended for instructors undergraduates and graduate students of mathematics physics chemistry and engineering Twenty eight chapters each prepared by an expert proceed from simpler to more difficult subjects Similarly the early chapters are easier than the later ones making the book ideal for independent study Subjects begin with the laws of thermodynamics and statistical theory of information and of ensembles advancing to the ideal classical gases of polyatomic molecules non electrolyte liquids and solutions and surfaces Subsequent chapters explore imperfect classical and quantum gas phase transitions cooperative phenomena Green function methods the plasma transport in gases and metals Nyquist s theorem and its generalizations stochastic methods and many other topics

Thermodynamics Ryōgo Kubo,1968 **An Introduction to Thermodynamics for Engineering Technologists** James P. Todd,Herbert B. Ellis,1981 Journal of the American Chemical Society American Chemical Society,1910 Proceedings of the Society are included in v 1 59 1879 1937 **An Introductory Guide to EC Competition Law and Practice** Valentine Korah,1994 **Theoretical chemistry from the standpoint of Avogadro's rule & thermodynamics** Walther Nernst,1916 **An Introduction to General Thermodynamics. An Elementary Treatise on the Fundamental Principles and Their Simpler Applications** Henry Augustus Perkins,1912 *Host Bibliographic Record for Boundwith Item Barcode 30112112290801 and Others* ,1903 **Catalogue** University of Virginia,1925

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