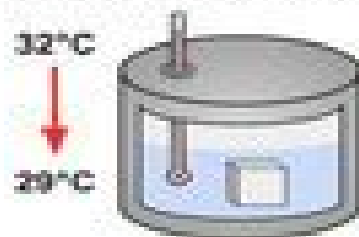


Instructions: Read each scenario and answer the questions below about calorimetry.



A piece of metal is placed in a calorimeter.

The temperature reading on the thermometer decreases from 32°C to 29°C .

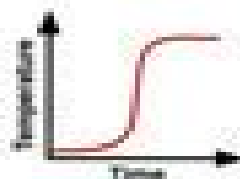
- 19) Which substance underwent an exothermic process? _____
- 20) Which substance underwent an endothermic process? _____
- 21) Heat flowed from the _____ into the _____

Calorimeter Readings	Temperature
Initial Temperature of Steel	150°C
Initial Temperature of Water	95°C
Final Temperature of Water	98°C

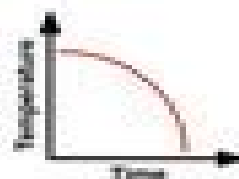
A cube of steel is placed in a calorimeter filled with water.

Initial and final temperatures are shown in the table.

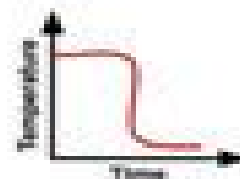
- 22) Which substance underwent an exothermic process? _____
- 23) Which substance underwent an endothermic process? _____
- 24) Which graph below shows the temperature of the steel? _____



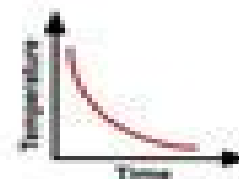
Graph A



Graph B



Graph C



Graph D

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Table of Contents Thermochemistry Study Guide With Heat Review

1. Understanding the eBook Thermochemistry Study Guide With Heat Review
 - The Rise of Digital Reading Thermochemistry Study Guide With Heat Review
 - Advantages of eBooks Over Traditional Books
2. Identifying Thermochemistry Study Guide With Heat Review
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an Thermochemistry Study Guide With Heat Review
 - User-Friendly Interface
4. Exploring eBook Recommendations from Thermochemistry Study Guide With Heat Review
 - Personalized Recommendations
 - Thermochemistry Study Guide With Heat Review User Reviews and Ratings
 - Thermochemistry Study Guide With Heat Review and Bestseller Lists
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 - Thermochemistry Study Guide With Heat Review Enhanced eBook Features
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 - Highlighting and Note-Taking Thermochemistry Study Guide With Heat Review
 - Interactive Elements Thermochemistry Study Guide With Heat Review
 8. Staying Engaged with Thermochemistry Study Guide With Heat Review
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 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Thermochemistry Study Guide With Heat Review
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 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Thermochemistry Study Guide With Heat Review
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Thermochemistry Study Guide With Heat Review
 - Setting Reading Goals Thermochemistry Study Guide With Heat Review
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Thermochemistry Study Guide With Heat Review
 - Fact-Checking eBook Content of Thermochemistry Study Guide With Heat Review
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning
 - Utilizing eBooks for Skill Development

- Exploring Educational eBooks

14. Embracing eBook Trends

- Integration of Multimedia Elements
- Interactive and Gamified eBooks

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