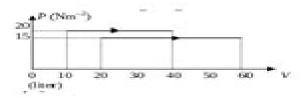
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April 23, 2018

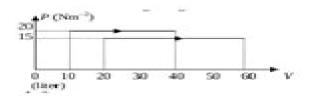


Thermodynamics - problems and solutions

The first law of thermodynamics

 Based on graph P-V below, what is the ratio of the work done by the gas in the process I, to the work done by the gas in the process II?

Known:



Process 1:

Pressure (P) = 20 N/m²

Initial volume (V₁) = 10 liter = 10 dm³ = 10×10^{-3} m³

Final volume $(V_2) = 40 \text{ liter} = 40 \text{ dm}^3 = 40 \times 10^3 \text{ m}^3$

Process 2:

Process (P) = 15 N/m²

Initial volume $(V_1) = 20 \text{ liter} = 20 \text{ dm}^3 = 20 \times 10^{-3} \text{ m}^3$

Final volume $(V_2) = 60 \text{ liter} = 60 \text{ dm}^3 = 60 \times 10^{-3} \text{ m}^3$

Wanted The ratio of the work done by gas

Solution:

Thermodynamics Problems And Solutions

Sanford Klein, Gregory Nellis

Thermodynamics Problems And Solutions:

Problems and Solutions on Thermodynamics and Statistical Mechanics Yung-kuo Lim, 1990 Volume 5

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