

MECHANICS OF FLUIDS

1. Determine the mass and the weight of the air contained in a room whose dimensions are 6 m x 6 m x 8 m. Assume the density of the air is 1.16 kg/m^3 .

1. Determined the mass and the weight of the air contained in a room whose dimensions are 6m x 6m x 8m. Assume the density of the air is 1.16 kg/m^3 .

Given :

$$\rho = 1.16 \text{ kg/m}^3 \longrightarrow \text{Density of air}$$

$$g = 9.81 \text{ m/s}^2 \longrightarrow \text{Assume gravity acceleration}$$

$$6\text{m} \times 6\text{m} \times 8\text{m} \longrightarrow \text{Dimension of room}$$

Solutions:

Volume of room:

$$V = 6\text{m} \times 6\text{m} \times 8\text{m} = 288 \text{ m}^3$$

Mass is defined by volume of room & air :

$$m = \rho \cdot V = \left(1.16 \frac{\text{kg}}{\text{m}^3}\right) (288 \text{ m}^3) = \boxed{334.08 \text{ kg}}$$

Weight of object :

$$G = m \cdot g = (334.08 \text{ kg}) (9.807 \frac{\text{m}}{\text{s}^2}) \left(\frac{1 \text{ N}}{1 \text{ kg} \cdot \frac{\text{m}}{\text{s}^2}}\right) = \boxed{3277.33 \text{ N}}$$

Therefore :

$$m = 334.08 \text{ kg}$$

$$G = 3277.33 \text{ N}$$

Armaline D. Quillas ~~Armaline D. Quillas~~ P.

Thermodynamics Problem And Solution

Ismail Tosun



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Thermodynamics Sanford Klein,Gregory Nellis,2011-10-10 This book differs from other thermodynamics texts

in its objective which is to provide engineers with the concepts tools and experience needed to solve practical real world energy problems The presentation integrates computer tools such as EES with thermodynamic concepts to allow engineering students and practising engineers to solve problems they would otherwise not be able to solve The use of examples solved and explained in detail and supported with property diagrams that are drawn to scale is ubiquitous in this textbook The examples are not trivial drill problems but rather complex and timely real world problems that are of interest by themselves As with the presentation the solutions to these examples are complete and do not skip steps Similarly the book includes numerous end of chapter problems both typeset and online Most of these problems are more detailed than those found in other thermodynamics textbooks The supplements include complete solutions to all exercises software downloads and additional content on selected topics These are available on the book s website www.cambridge.org KleinandNellis

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Thermodynamics Martin Dehli, Ernst Doering, Herbert Schedwill, 2022-10-29 This textbook presents the essential scientific principles of thermodynamics in a detailed and well structured manner for practice oriented teaching It conveys analytically reliable knowledge with a view to engineering application and provides the key to a quick understanding of e g thermal machines heat transfer humid air and combustion The present English edition in comparison to earlier German editions has been extended to include aspects of fluid mechanics dynamics of ideal gases and chemical thermodynamics **The**

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Ruskell, 2008 Thermodynamics Stephen R. Turns, Laura L. Pauley, 2020-02-27 Fully revised to match the more traditional sequence of course materials this full color second edition presents the basic principles and methods of thermodynamics using a clear and engaging style and a wealth of end of chapter problems It includes five new chapters on topics such as mixtures psychometry chemical equilibrium and combustion and discussion of the Second Law of Thermodynamics has been expanded and divided into two chapters allowing instructors to introduce the topic using either the cycle analysis in Chapter 6 or the definition of entropy in Chapter 7 Online ancillaries including new LMS testbanks a password protected solutions manual prepared PowerPoint lecture slides instructional videos and figures in electronic format are available at www.cambridge.org/thermo

A Problem-Solving Approach to Aquatic Chemistry James N. Jensen, 2023-01-05 A Problem Solving Approach to Aquatic Chemistry Enables civil and environmental engineers to understand the theory and application of aquatic equilibrium chemistry The second edition of A Problem Solving Approach to Aquatic Chemistry provides a detailed introduction to aquatic equilibrium chemistry calculation methods for systems at equilibrium applications of aquatic chemistry and chemical kinetics The text directly addresses two required ABET program outcomes in environmental engineering chemistry including stoichiometry equilibrium and kinetics and material and energy balances fate and transport of substances in and between air water and soil phases The book is very student centered with each chapter beginning with an introduction and ending with a summary that reviews the chapter s main points To aid in reader comprehension important terms are defined in context and key ideas are summarized Many thought provoking discussion questions worked examples and end of chapter problems are also included Each part of the text begins with a case study a portion of which is addressed

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