

General Chemistry placement test Study Guide Exam with 100% Verified Solutions

molecule - answer✓✓two nonmetal atoms combined form a covalent compound called a

diatomic molecules - answer✓✓H₂,O₂,F₂

Matter - answer✓✓Anything that has mass and takes up space

Element - answer✓✓A substance that cannot be broken down into simpler substances

Compound - answer✓✓A pure substance made of two or more elements **CHEMICALLY** combined.

Compound - answer✓✓Can be decomposed by chemically

Mixture - answer✓✓A **PHYSICAL** blend of two or more components

(aq) - answer✓✓aqueous mixture, a substance is dissolved in water

homogeneous mixture - answer✓✓A mixture that is so evenly mixed that you can't see all of the different parts. Ex. Air, kool-aid

heterogeneous mixture - answer✓✓A mixture in which different materials can be seen easily.

Ex. soil, chocolate chip cookies

filtration - answer✓✓process to separate a solid "precipitate" from a liquid

sublimation - answer✓✓phase change from (s) → (g)

empirical formula - answer✓✓A formula showing the lowest whole number ratio of atoms in a compound. Can't be reduced.

solid - answer✓✓A state of matter that has a definite shape and a definite volume

liquid - answer✓✓A form of matter that flows, has constant volume, and takes the shape of its container

Bromine (Br) and Mercury (Hg) - answer✓✓only two liquids on the periodic table

gas - answer✓✓A state of matter with no definite shape or volume

gas - answer✓✓will uniformly and completely fill a container

Wayne State Chemistry Placement Exam Study Guide

Mark Steyvers



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