



Solubility Rules

Rule #		SOLUBILITY RULES
1	Cations	ALL Group 1 alkali metal (Li^+ , Na^+ , K^+ , Rb^+ , Cs^+ , Fr^+) and Ammonium (NH_4^+) compounds are <i>soluble</i> in water. Most, but NOT all, Group 2 alkali earth metal compounds are <i>soluble</i> .
2	Halides	Most Halides are <i>soluble</i> . Chlorides (Cl^-), Bromides (Br^-), and Iodides (I^-) are <i>soluble</i> EXCEPT Silver (Ag^+), Mercury (Hg_2^{2+}), and Lead (Pb^{2+})
3	Soluble Anions	Nitrates (NO_3^-), Acetates ($\text{C}_2\text{H}_3\text{O}_2^-$), Chlorate (ClO_3^-), and Perchlorates (ClO_4^-) are <i>soluble</i>
4	Sulfates	Most Sulfates (SO_4^{2-}) are <i>soluble</i> EXCEPT compounds of Barium (Ba^{2+}), Strontium (Sr^{2+}), Calcium (Ca^{2+}), Lead (Pb^{2+})
5	Insoluble Anions	Most Phosphates (PO_4^{3-}), Carbonates (CO_3^{2-}), Sulfides (S^{2-}), & Sulfites (SO_3^{2-}) are <i>insoluble</i> EXCEPT for compounds of Group 1 alkali metals and Ammonium (NH_4^+)
6	Hydroxides	Most hydroxides (OH^-) are <i>insoluble</i> EXCEPT compounds of Group 1 alkali metals, most Group 2 alkali earth metals, and Ammonium (NH_4^+)

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In a global consumed by monitors and the ceaseless chatter of instantaneous communication, the melodic beauty and emotional symphony developed by the published term often diminish into the backdrop, eclipsed by the relentless sound and interruptions that permeate our lives. Nevertheless, nestled within the pages of **Ws15 1 Solubility Rules** a marvelous fictional value overflowing with organic emotions, lies an immersive symphony waiting to be embraced. Constructed by a masterful composer of language, this captivating masterpiece conducts readers on a psychological journey, skillfully unraveling the concealed songs and profound influence resonating within each cautiously constructed phrase. Within the depths of this emotional review, we can examine the book is central harmonies, analyze their enthralling writing style, and surrender ourselves to the profound resonance that echoes in the depths of readers souls.

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