

# Unit 4.8 STOICHIOMETRY part 2

(Show all work using dimensional analysis)

1.  $2 \text{ sugar-beets} + 1 \text{ egg} \rightarrow 1 \text{ soufflé}$

a) If you have 12 cups of flour and 3 eggs, how many soufflés of maximum you make?

Ans: \_\_\_\_\_

b) What ingredient do you run out of first (limiting reagent)?

Ans: \_\_\_\_\_

2.  $2 \text{ Al} + 6 \text{ HCl} \rightarrow 2 \text{ H}_2 + 2 \text{ AlCl}_3$

a) If you start with 200 g of aluminum and 750 g of HCl, how many grams of  $\text{H}_2$  can be made?

Ans: \_\_\_\_\_

b) What is the limiting reagent?

Ans: \_\_\_\_\_

c) Suppose you only make 100 g of  $\text{H}_2$ . What is your % yield?

Ans: \_\_\_\_\_

3.  $2 \text{ Fe}_2\text{O}_3 + 3 \text{ C} \rightarrow 4 \text{ Fe} + 3 \text{ CO}_2$

a) How many grams of iron can be made from 100 g of  $\text{Fe}_2\text{O}_3$  and 12.7 g of C?

Ans: \_\_\_\_\_

b) What is the limiting reagent?

Ans: \_\_\_\_\_

c) After the above reaction, you produce 25.0 g of Fe. What is your % yield?

Ans: \_\_\_\_\_

4.  $6 \text{ CO}_2 + 6 \text{ H}_2\text{O} \rightarrow \text{C}_6\text{H}_{12}\text{O}_6 + 6 \text{ O}_2$

a) How many grams of sugar can be made from 500 g of  $\text{CO}_2$  and 500 g of water?

Ans: \_\_\_\_\_

b) What is the limiting reagent?

Ans: \_\_\_\_\_

c) What is the excess reagent?

Ans: \_\_\_\_\_

**BONUS:** How many grams of three extra reagent are leftover?

How much of the reagent left over?

Ans: \_\_\_\_\_

0.000 0.001 0.01 0.1 1 10 100 1000 10000 100000 1000000

1000000 100000 10000 1000 100 10 1 0.1 0.01 0.001 0.000

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**Yicheng Fang**



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