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web define isotope using mass number atomic number number of protons neutrons and electrons given information about an element find the mass and name of an isotope give evidence to support or dispute in nature the chance of finding one isotope

of an element is the same for all isotopes  $% \frac{1}{2}\left( \frac{1}{2}\right) =\frac{1}{2}\left( \frac{1}{2}\right) =\frac{1}$ 

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web the atomic mass of a single atom is simply its total mass and is typically expressed in atomic mass units or amu by definition an atom of carbon with six neutrons carbon 12 has an atomic mass of 12 amu other atoms don t generally have round number atomic masses for reasons that are a little beyond the scope of this article chapter 1 7 the mole and molar mass chemistry libretexts - Jul 02 2022

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