



Chapter Review

Properties of Atoms and the Periodic Table

Part A. Vocabulary Review

Directions: Use the space at the end of each word to write the correct definition of the word as you understand it. Do not copy the definition.

metals (physical properties)	halogens (chemical properties)	average atomic mass (mass of atoms, elements)	electron's orbital (atomic structure)
metalloids	metals	mass number	periods
periodic system	quarks	transition metals	transition metals
_____	1. Its original form is _____ and original form and a small form that is used as a reference in science is called _____.	_____	_____
_____	2. The transition metals are elements that are _____.	_____	_____
_____	3. All groups of elements in the periodic table have a similar chemical property. The element in the _____.	_____	_____
_____	4. The most abundant elements are called _____.	_____	_____
_____	5. Elements in the middle of the periodic table group are called _____.	_____	_____
_____	6. The transition metals are elements that _____.	_____	_____
_____	7. Elements that are in the middle of the periodic table are called _____.	_____	_____
_____	8. The elements that are in the middle of the periodic table are called _____.	_____	_____
_____	9. The elements that are in the middle of the periodic table are called _____.	_____	_____
_____	10. The elements that are in the middle of the periodic table are called _____.	_____	_____
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_____	19. The elements that are in the middle of the periodic table are called _____.	_____	_____
_____	20. The elements that are in the middle of the periodic table are called _____.	_____	_____

Vocabulary Review For Understanding The Atoms Guide

George Graybill



Vocabulary Review For Understanding The Atoms Guide:

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