

Chapter 2: Water

2.1 Water Weak Interactions in Aqueous Systems

Water's Qualities:

-Unusual properties of water caused by Hydrogen Bonds:

- 1) High Surface tension – beads up
- 2) High heat capacity
- 3) High Heat of Vaporization – a lot of energy is needed for water to boil
- 4) Expands as it freezes – the solid forms of water (ice) floats instead of sinks like most

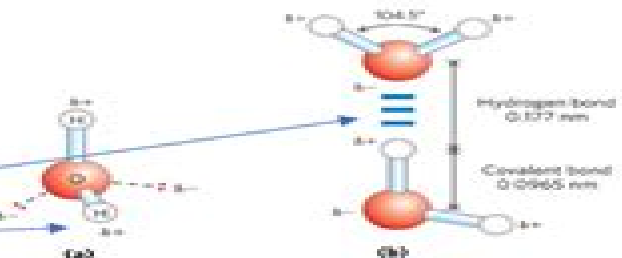
Inter/Intramolecular Forces 1-HYDROGEN BONDING

-Hydrogen bonding occurs when the hydrogen atoms of one molecule is attracted to another molecule that has Nitrogen, Oxygen, and Sulfur (NOS).

- Oxygen is the most electronegative followed by sulfur and Nitrogen.

-Picture to the right: Water Molecules

- O is more electronegative than H
- This creates a POLAR covalent bond between H and O
- The H-Bond occurs between water molecules
- O has a partial negative (-) charge
- H has a partial positive (+) charge



-Polar function groups form H-Bonds with one another and water



-Angle and distance affect the strength of H-Bonds



-we will see H-Bonds occur between:

- Hydroxyl group of alcohol and water
- Carbonyl group of a ketone and water
- between peptide groups and peptides
- complementary bases of DNA

Between the hydroxyl group of an alcohol and water



Between the carbonyl group of a ketone and water



Between peptide groups in polypeptides



Between complementary bases of DNA



Water And Aqueous Systems Study Guide

Pandey S.N. & Sinha B.K.



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Study Guide for Atoms, Molecules, and Life Michael S. Matta, 1981 *The Absolute, Ultimate Guide to Lehninger Principles of Biochemistry* Marcy Osgood, Karen A. Ocorr, 2008-04-04 The Absolute Ultimate Guide combines an innovative study guide with a reliable solutions manual in one convenient printed volume **Guide to the Analysis of Pesticide Residues** Harry Phineas Burchfield, Donald E. Johnson, 1965 Water Quality Instructional Resources Information System (IRIS) , 1979 **Indexes** United States. Environmental Protection Agency, 1983 **Ground Water Models** National Research Council, Division on Engineering and Physical Sciences, Commission on Physical Sciences, Mathematics, and Applications, Committee on Ground Water Modeling Assessment, 1990-02-01 The discovery of toxic pollution at Love Canal brought ground water contamination to the forefront of public attention Since then ground water science and modeling have become increasingly important in evaluating contamination setting regulations and resolving liability issues in court A clearly written explanation of ground water processes and modeling Ground Water Models focuses on the practical aspects of model application It examines the role of models in regulation litigation and policy development explains ground water processes and describes specific applications for models presents emerging technologies and offers specific recommendations for better use of ground water science in policy formation *EPA-430/1* , 1979-05 **Student Study Guide to accompany Chemistry** Martin Silberberg, Libby Weberg, 2005-01-06 *Water Resources in Mexico* Úrsula Oswald Spring, 2011-08-26 Water resources in Mexico are threatened by scarcity pollution and climate change In two decades water consumption doubled producing water stress in dry seasons and semi arid and arid regions Water stress rises due to physical and economic stress In seven parts a multidisciplinary team analyzes hydrological processes in basins and their interaction with climate soil and biota Competing water use in agriculture industry and domestic needs require savings decontamination processes and desalination to satisfy the growing demand Water quality affects health and ecosystems This creates conflicts and cooperation that may be enhanced by public policy institution building and social organization *Selected Water Resources Abstracts* , 1990 Research and Technology Program Digest United States. National Aeronautics and Space Administration, *Thermodynamics and Kinetics of Water-Rock Interaction* Eric H. Oelkers, Jacques Schott, 2018-12-17 Volume 70 of Reviews in Mineralogy and Geochemistry represents an extensive review of the material presented by the invited speakers at a short course on Thermodynamics and Kinetics of Water Rock Interaction held prior to the 19th annual V M Goldschmidt Conference in Davos Switzerland June 19 21 2009 Contents Thermodynamic Databases for Water Rock Interaction Thermodynamics of Solid Solution Aqueous Solution Systems Mineral Replacement Reactions Thermodynamic Concepts in Modeling Sorption at the Mineral Water Interface Surface Complexation Modeling Mineral Fluid Equilibria at the Molecular Scale The Link Between Mineral Dissolution Precipitation Kinetics and Solution Chemistry Organics in Water Rock Interactions Mineral Precipitation Kinetics Towards an Integrated Model of Weathering Climate and Biospheric Processes

Approaches to Modeling Weathered Regolith Fluid Rock Interaction A Reactive Transport Approach Geochemical Modeling of Reaction Paths and Geochemical Reaction Networks *Study Questions and Problems in Inorganic Chemistry* Alexander Silverman,1924 **Plant Physiology, 4th Edition** Pandey S.N. & Sinha B.K.,2009-11-01 Plant Physiology lucidly explains the operational mechanisms of plants based on up to date literature and with the help of numerous illustrations In addition to the theoretical aspects experiments have been incorporated at the end of relevant chapters The book with its compilations of vast literature and its lucid presentation will certainly be useful to undergraduate and postgraduate students It will also be of help to students preparing for various competitions including IAS PCS and Medical Entrance Examinations of various boards

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