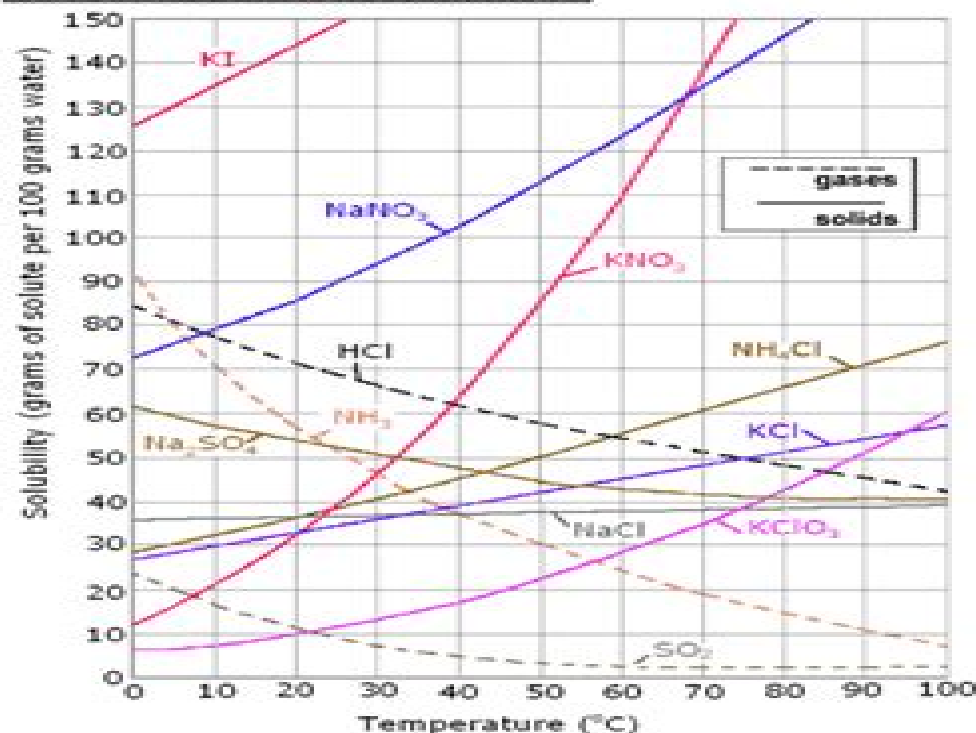


## UNIT 12 - SOLUTIONS

### SOLUBILITY CURVES WORKSHEET



- Which compound is *least* soluble at 20 °C? At 80 °C?
- Which substance is the *most* soluble at 10 °C? At 50 °C? At 90 °C?
- The solubility of which substance is *most* affected by changes in temperature?
- The solubility of which substance is *least* affected by changes in temperature?
- Are the following solutions saturated, unsaturated, or supersaturated?  
 (Assume all are dissolved in 100 grams of water.)  
 (A) 50 grams of KNO<sub>3</sub> at 50 °C  
 (B) 100 grams of NaNO<sub>3</sub> at 80 °C  
 (C) 30 grams of KNO<sub>3</sub> at 25 °C  
 (D) 50 grams of KCl at 80 °C  
 (E) 65 grams of NH<sub>4</sub>Cl at 70 °C  
 (F) 90 grams of KNO<sub>3</sub> at 60 °C
- NH<sub>3</sub> is a gas. Describe what happens to its solubility as the temperature goes from 20 °C to 80 °C.

# Unit 12 Solutions Solubility Curves

**Wolfgang Guggemos**

A red circular graphic with a gradient, appearing as a partial circle or a stylized arrow pointing to the right, located to the right of the author's name.

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