Understanding the Periodic Table

 Consolidate your understanding of the periodic table by using the word bank to complete the blanks below.

Word bank: periods protons noble gases alkaline earth metals groups synthetic atomic number halogens electrons bromine alkali metals metals mercury

Most elements in the Periodic Table occur naturally; however, some, such as technetium (Tc), rutherfordium (Rf), and fermium (Fm), are ______, meaning they were created in the laboratory.

Some groups in the Periodic Table have special names. Group 1 elements are known as the _______, Group 2 elements are called the _______, Group 17 elements are the _______, and Group 18 elements are the noble gases. Elements in the same group have the same number of _______ in their outer shell, which means they have similar chemical properties and tend to react in similar ways.

The ______ of an element increases from left to right across the Periodic Table. The atomic number represents the number of _____ present in the nucleus of an atom of that element. Each element has a unique atomic number.

Most elements in the Periodic Table are ______, some are nonmetals, and a few are metalloids. Metalloids are elements that display properties of both metals and nonmetals.

(Discrevelbreik)

Unscrambling The Periodic Table Answers

A Loxley

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|---|
| example the chemical word POSSEsSiON is made using the symbols for phosphorus P oxygen O sulfur S Einsteinium Es |
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