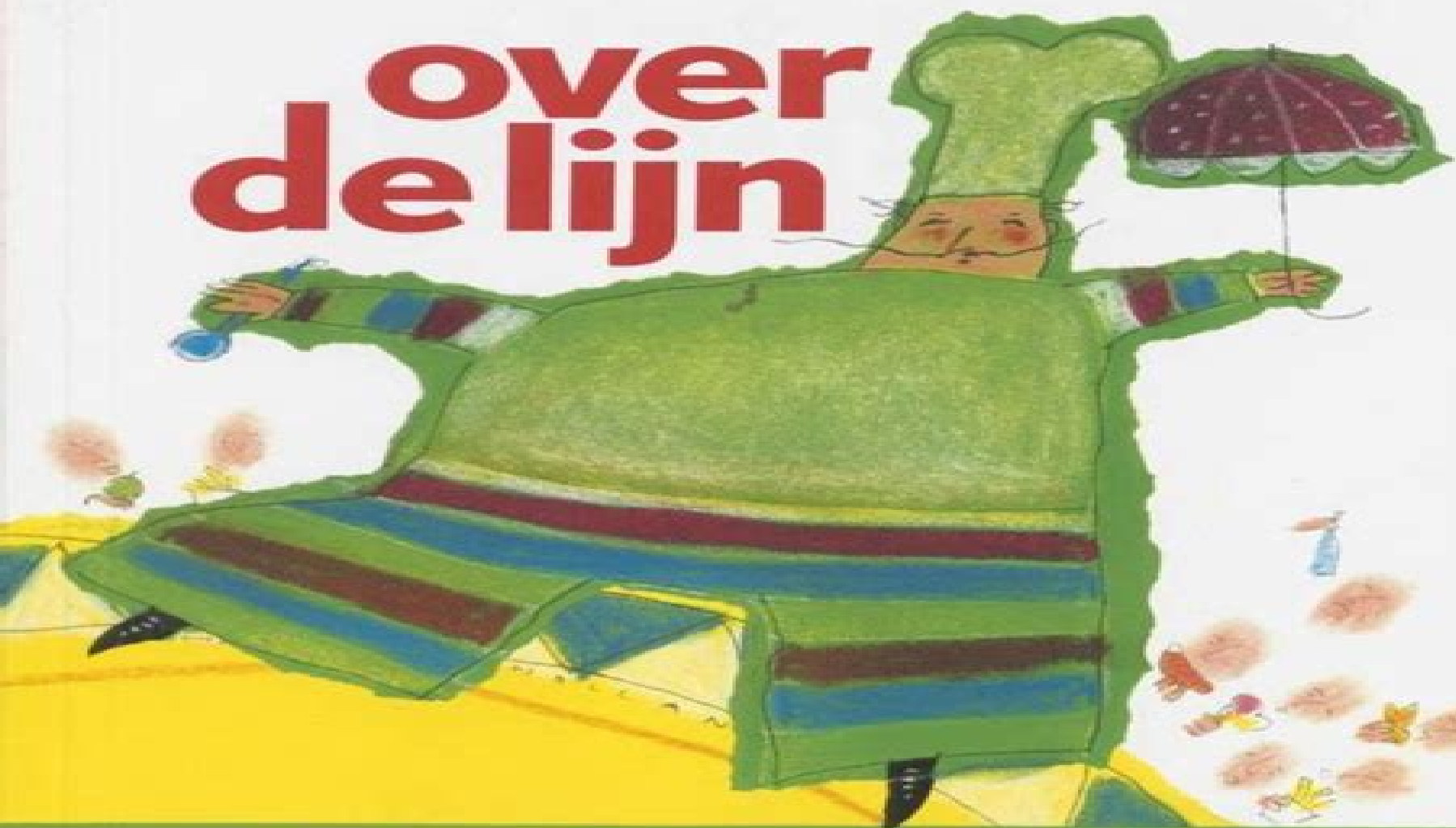


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Zin En Onzin Van Dieten Over De Lijn Tekeningen Dick Valentijn

Huangqi Zhang



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web in each of the following redox reactions indicate the elements that undergo changes in oxidation number and the magnitude of change then identify the element oxidized and reduced as well as the oxidizing agent and reducing agent
 $\text{I}_2\text{O}_5 + 5\text{CO} \rightarrow \text{I}_2 + 5\text{CO}_2$

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 $2\text{Sr} + \text{O}_2 \rightarrow 2\text{SrO}$ Sr^0 to Sr^{+2} oxidized reducing agent O^0 to O^{+2} reduced
 $\text{Ox ag } 2\text{Li} + \text{S} \rightarrow \text{Li}_2\text{S}$ Li^0 to Li^{+1} oxidized red ag S^0 to S^{+2} reduced
 $\text{Ox ag } 3\text{C} + 2\text{Br}_2 \rightarrow \text{C}_2\text{Br}_4$

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